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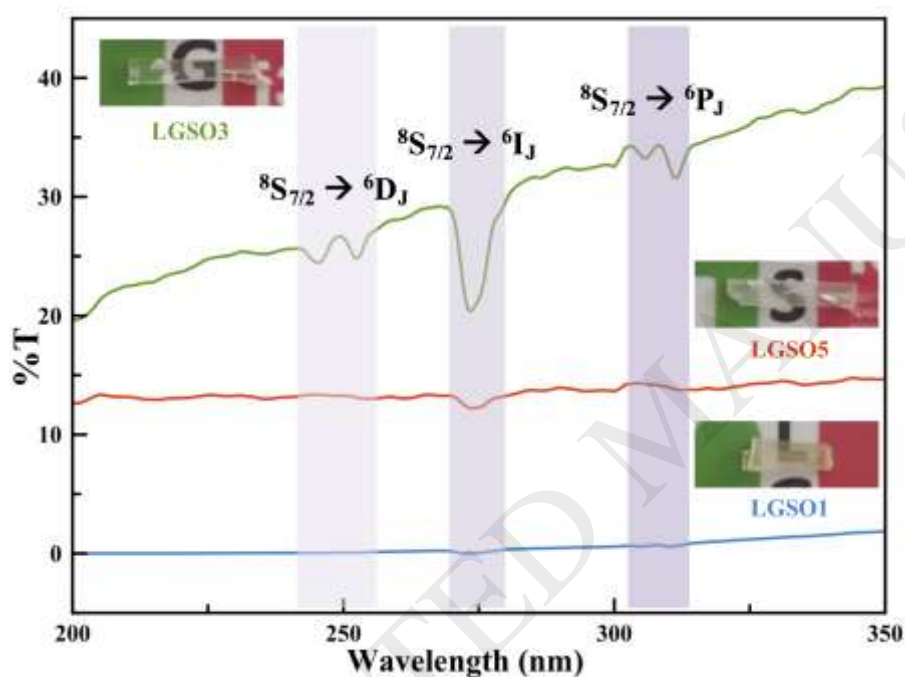
# Influence of Lu content on $(\text{Lu}_x\text{Gd}_{1-x})_2\text{SiO}_5$ oxyorthosilicates grown by Laser Floating Zone: structural studies and transparency

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## GRAPHICAL ABSTRACT



## HIGHLIGHTS

- Lutetium gadolinium oxyorthosilicates (LGSO) produced by Laser Floating Zone (LFZ).
- High growth rate in air of LGSO ( $(\text{Lu}_x\text{Gd}_{1-x})_2\text{SiO}_5$ ,  $x=0.1, 0.3$  and  $0.5$ ) single crystal.

- Transparency and energy transfer band intensity tailored by the Lu/Gd amount.

## ABSTRACT

Lutetium gadolinium oxyorthosilicates (LGSO,  $(\text{Lu}_x\text{Gd}_{1-x})_2\text{SiO}_5$ ,  $x=0.1, 0.3$  and  $0.5$ ) were obtained by the Laser Floating Zone (LFZ) in air at 10 mm/h, much faster than those produced by the standard Czochralski method. The LGSO fibres were structural and optically characterized. Scanning electron microscopy (SEM/EDS) allowed observing homogeneous monophasic crystalline fibres that obey the expected phase transition from  $P2_1/c$  to  $C2/c$  monoclinic structures, as revealed by X-Ray diffraction and Raman spectroscopy. Finally, transmission studies in the near UV to visible range allowed to quantify the transmission and to appraise the presence of energy transfer bands that favor their use as host materials in photonic applications.

**Keywords:** LGSO, Oxyorthosilicates, Crystal growth, Phase transition, Laser Floating Zone

## INTRODUCTION

Gadolinium (GSO) and lutetium (LSO) oxyorthosilicates have been studied in different photonics **applications** [1-4], however the interest of researchers has progressively been directed towards mixed oxyorthosilicates of lutetium and gadolinium (LGSO) [5-8] due to three important reasons [9-11].

Lutetium, as pure  $\text{Lu}_2\text{O}_3$  precursor, is an expensive material that significantly increases production costs when it is employed on photonic materials production. Thus, co-doping with gadolinium ions from, for example, cheaper  $\text{Gd}_2\text{O}_3$ , allows reducing production associated costs without affecting the structural and optical properties. On the other hand, gadolinium oxyorthosilicate single crystals **tend** to develop cracks during growth. In this case, lutetium doping has been employed **aiming to** reduce thermal stress and, therefore, avoid crack formation. Finally, compared to pure LSO materials, the gadolinium doping significantly reduces the melting point.

Since Loutts et al. [12] produced LGSO for first time in 1997, to our knowledge, most studies related to its synthesis involve the use of the Czochralski (CZ) method [5-12]. Only Zorenko et al. [13, 14] reported other technique such as liquid phase epitaxy for the growth of single crystalline films. **To our knowledge, De la Fuente et al. [15] produced for first time Nd-doped fibres of LSO and GSO in Ar+O<sub>2</sub> atmosphere by the alternative Laser Floating Zone (LFZ) technique.** Recently, pure GSO [16] and LSO [17] have been obtained for first time in air at high pulling rates employing **this technique [16-17]**. However, the growth behavior of these oxyorthosilicates under the LFZ processing is quite different [18]. Thus, while single crystalline monoclinic fibres of  $\text{Gd}_2\text{SiO}_5$  can be obtained in air at 10 mm/h [16] as expected from crystal growth theory [19], eutectic fibres of LSO/ $\text{Lu}_2\text{O}_3$  are obtained for the  $\text{Lu}_2\text{O}_3:\text{SiO}_2$  system **at the same pulling rate [17]**. **This contradictory behavior relates to the high difference between precursors melting points [17, 18], specifically considering each oxide type structure [20]. Gadolinium oxide starts a structural transition from cubic (Ia $\bar{3}$ ) C-type to monoclinic (C2/m) B-type at 1500 °C. However, lutetium oxide only presents the cubic (Ia $\bar{3}$ ) C-type structure, remaining thermally stable until 2400 °C [21], well above the silicon oxide boiling point (2230 °C). Thus, silicon oxide can be evaporated before reacting with molten lutetium oxide.** This fact, together with the interest of **improving the** structural and optical properties of GSO fibres produced by LFZ **and the possibility to avoid that effect**, justifies a study on gadolinium lutetium oxyorthosilicates developed by this technique.

Thus, the aim of this work was the study of the Lu doping influence on the structural and physical properties of LGSO materials processed by the LFZ technique. Compositions of  $(\text{Lu}_x\text{Gd}_{1-x})_2\text{SiO}_5$  (LGSO) oxyorthosilicate, with  $x= 0.1, 0.3$  and  $0.5$ , were studied. The morphology and structural characteristics of the LGSO fibres produced were assessed by scanning electron microscopy

(SEM/EDS), Raman spectroscopy and X-ray diffraction (XRD) analysis. Finally, transmission studies on UV-Vis range were carried out to confirm if optical properties associated to intra  $4d^{f7}$  transitions of  $Gd^{3+}$  ions related to energy transfer band that allows their use as laser host material are kept. The transparency was evaluated by absorption experiments in the **visible - near UV region**.

## EXPERIMENTAL

### Crystal growth

$(Lu_xGd_{1-x})_2SiO_5$  (LGSO) oxyorthosilicate fibres, with  $x = 0.1, 0.3$  and  $0.5$ , were grown by the LFZ technique [16, 17, 22]. Feed and seed rod precursors for LFZ growth were prepared by mixing  $Gd_2O_3$  (Mateck, 99.999%),  $Lu_2O_3$  (Shangai Co., 99.99%) and  $SiO_2$  (Aldrich, 99.99%) in agate ball mill commercial powders at the established proportions (Table I). Polyvinyl alcohol (PVA) was added to bind the powder mixture (PVA 0.1 g/ml) that was further extruded into cylindrical rods with diameters around 1.5 mm.

The LFZ equipment comprises a 200 W  $CO_2$  laser (Spectron, GSI group) coupled to a reflective optical setup producing a circular crown-shaped laser beam promoting a uniform radial heating [18, 22]. This way, fibres with diameters around 1.5 mm and 10-40 mm in length were grown in descendent direction at 10 mm/h pulling rate on air at atmospheric pressure under a rotation speed of 5 rpm **from rods previously densified at 100 mm/h [16, 17]. Given the densified nature of the feed rods, both feed and seed rates were the same, 10mm/h.** Table I summarizes the studied compositions and corresponding growth conditions. **Those of pure GSO and LSO ( $x = 0$  and  $x = 1$ , respectively) previously grown by LFZ in the same conditions [16, 17], are also included for reference.**

**Table I. Composition and LFZ experimental conditions, compared to those of pure GSO [16] and LSO [17].**

Sample	Nominal formula	Diameter (mm)	Pulling rate (mm/h)	Power (W)	Ref.
GSO	Gd <sub>2</sub> SiO <sub>5</sub>	1.5	10	72	<b>16</b>
LGSO-1	(Lu <sub>0.1</sub> Gd <sub>0.9</sub> ) <sub>2</sub> SiO <sub>5</sub>	1.5	10	67	-
LGSO-3	(Lu <sub>0.3</sub> Gd <sub>0.7</sub> ) <sub>2</sub> SiO <sub>5</sub>	1.5	10	58	-
LGSO-5	(Lu <sub>0.5</sub> Gd <sub>0.5</sub> ) <sub>2</sub> SiO <sub>5</sub>	1.5	10	64	-
LSO	Lu <sub>2</sub> SiO <sub>5</sub>	2	10	92	<b>17</b>

### Structural and Optical characterization

The fibres microstructure was characterized by scanning electron microscopy (SEM, TESCAN Vega 3SEM) fitted with energy dispersive X-ray spectroscopy (EDS) on polished surfaces of longitudinal and transversal fibre sections.

The structural characterization of the fibres was firstly accomplished by X-ray powder diffraction (XRD) experiments with a EMPYREAN X-ray diffractometer (PANalytical) using a Cu anode, while the data were fitted with the analytical software HighScore Plus. The fibres were smashed into powder to perform the standard Bragg-Brentano X-ray diffraction analysis. Room temperature (RT)  $\mu$ -Raman spectroscopy using the 441.6 nm line of a He–Cd laser (Kimmon IK Series) was also performed to complete structural characterization. A Horiba Jobin-Yvon HR800 instrument fitted with a 100x magnification lens with 0.9 numerical aperture and a minimum spot size < 2  $\mu$ m was employed to perform the Raman studies.

The transmission measurements were carried out in a Shimadzu UV-2100 spectrometer which was used in the range of 190-900 nm with deuterium and tungsten lamps as excitation sources. Fibres polished on plane-parallel configuration were employed as measuring samples.

## RESULTS AND DISCUSSION

Lutetium-gadolinium oxyorthosilicate crystalline fibres have been successfully obtained by LFZ, as shown in Fig. 1, where plane parallel-polished fragments of materials developed are shown. When

observed at naked eye, samples differ on transparency, being both LGSO-1 and LGSO-5 fibres translucent. In its turn, the LGSO-3 sample is transparent, being clearly distinguishable the colors of the image background and displaying higher sharpness. Additionally, the LGSO-1 sample presents a yellowish tone **like** that observed in GSO samples [16]. In fact, its composition is the closest to the pure gadolinium oxyorthosilicate, for which the Lu content is minimum. On the other hand, the LGSO-5 seems to be neutral translucent allowing us to consider that the LSO behavior could be achieved at low pulling rates for  $\text{Lu}_2\text{O}_3\text{-SiO}_2$  systems in lutetium oxyorthosilicates with high amounts of Gd in opposite to **not doped** systems processed by LFZ that produced ceramic biphasic samples [17]. Comparing to the **pure** GSO processed by LFZ [16], the LGSO-3 fibre exhibits a lower amount of internal cracks and fractures. This corroborates the bibliography, since LGSO possesses better plasticity and lower tendency to crack than **pure** GSO [5, 9-11]. The cracks observed in Fig. 1 are mainly originated from the cutting and polishing steps. These first results together with the advantages promoted by LFZ technique such as no seed needed or processing under ambient conditions should drastically reduce fabrication costs favoring its value over standard methods like CZ. Nevertheless, the structure and compositions must be studied in detail.

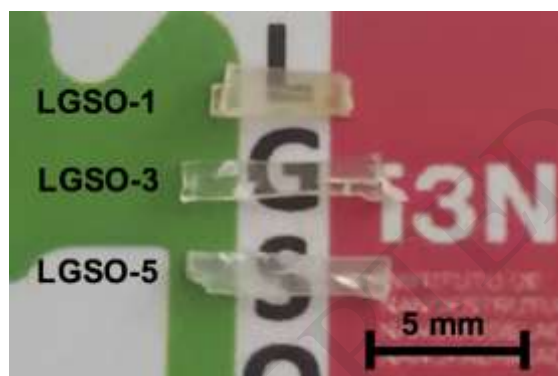


Fig. 1 Photograph of bulk plane parallel-polished fragments of LGSO-1 (top), LGSO-3 (middle) and LGSO-5 (bottom) samples grown at 10 mm/h.

SEM analysis of the fibres grown by LFZ reveals that the longitudinal sections of all LGSO samples (Fig. 2) are homogeneous along the entire surface with uniform contrast and without visible grain boundaries, suggesting **single crystal** [16, 17]. Moreover, no significant cracks due to **stresses** are

observed at the border of the fibres (Fig. 2b). However, small cracks and internal defects probably due to quick crystallization and/or polishing processes are observed inside the fibres. Considering that the LGSO-5 fibre presents the major **number** of defects and that all the fibres were polished together, it **could** be assumed that this sample corresponds to the weaker material. So, the introduction of the Lu could induce a reduction on the structural stability of the LGSO fibres grown by LFZ. Accordingly, Dominiak-Dzik et al. [9], observed how  $\text{Lu}^{3+}$  substitution by  $\text{Gd}^{3+}$  ions on  $(\text{Lu}_{1-x}\text{Gd}_x)_2\text{SiO}_5:\text{Dy}$  single crystals grown by CZ decreases melting point, reducing stress and, therefore, possible cracks on oxyorthosilicate structure. **However, observing the samples appearance (Fig. 1), seems to be a non-monotonic behavior with the Lu doping. Indeed, while LGSO-1 and LGSO-5 present similar transparency, the LGSO-3 sample is the best of the set. Thus, structure and composition were studied in detail to understand its influence on optical properties.**



Fig. 2 Longitudinal section SEM micrographs (500X) of (a) LGSO-1, (b) LGSO-3 and (c) LGSO-5 fibres.

On the other hand, EDS analysis performed confirms the homogeneity of fibres, revealing a uniform composition without presence of secondary or eutectic phases, as could be expected from the LFZ treatment of  $\text{Lu}_2\text{O}_3/\text{SiO}_2$  mixtures [17]. It must be noted that compositions calculated from EDS analysis are close to initial mixtures (Table II), **in opposition** to compositional dissimilarity observed for LGSO materials developed by conventional CZ method [5-12]. Anyway, **aiming to** reduce fabrication costs associated to LSO materials by standard methods like CZ, **LFZ results more** suitable from the environmental point of view (growth in ambient conditions) and time reduction. Moreover, the introduction of dopants avoids the formation of eutectic phases as Rey-Garcia et al. [17] reported for the LFZ processing of the  $\text{Lu}_2\text{O}_3/\text{SiO}_2$  system at 10 mm/h.

**Table II.** Initial and calculated compositions determined by EDS analysis.



Sample	Initial composition	EDS calculated composition
LGSO-1	(Lu <sub>0.1</sub> Gd <sub>0.9</sub> ) <sub>2</sub> SiO <sub>5</sub>	(Lu <sub>0.12</sub> Gd <sub>0.88</sub> ) <sub>2</sub> SiO <sub>5</sub>
LGSO-3	(Lu <sub>0.3</sub> Gd <sub>0.7</sub> ) <sub>2</sub> SiO <sub>5</sub>	(Lu <sub>0.31</sub> Gd <sub>0.69</sub> ) <sub>2</sub> SiO <sub>5</sub>
LGSO-5	(Lu <sub>0.5</sub> Gd <sub>0.5</sub> ) <sub>2</sub> SiO <sub>5</sub>	(Lu <sub>0.53</sub> Gd <sub>0.47</sub> ) <sub>2</sub> SiO <sub>5</sub>

XRD analysis of the LGSO powders puts in evidence the crystallinity degree and the monophasic nature of these samples together with the **expected** structural transition. The diffractograms recorded for LGSO sample were compared to those of un-doped GSO [16], LSO [17] and corresponding XRD ICDD cards [23], 04-009-2670 and 04-013-1146, respectively, to achieve a good understanding. As we can observe on Fig. 3, LGSO-1 presents the monoclinic P2<sub>1</sub>/c structure while the LGSO-3 and LGSO-5 have a monoclinic C2/c structure. These results follow the structural phase transition observed for (Lu<sub>x</sub>Gd<sub>1-x</sub>)<sub>2</sub>SiO<sub>5</sub> (LGSO) oxyorthosilicate materials that exhibit a GSO-type (P2<sub>1</sub>/c) structure for x<0.15 and LSO-type (C2/c) when x>0.17 [5-14].

The phase transition rule can be explained from the different atomic size together with the ion surrounding. Recently, Ryba-Romanowski et al. [6] related the crystal transition of LGSO doped compositions to the differences of nearest surrounding rare earth ions. Two types of Gd sites, Gd1 and Gd2, are present at the GSO lattice, presenting both different coordination number and local symmetries (CN=9, C<sub>3v</sub> and CN=7, C<sub>s</sub>, respectively). These promote the polyhedrons GdO<sub>9</sub> and GdO<sub>7</sub> depicted in Fig. 4. However, in the case of the lutetium oxyorthosilicate, both lutetium ions and, therefore, the LuO<sub>6</sub> (Lu1) and LuO<sub>7</sub> (Lu2) polyhedrons, present the C<sub>s</sub> local symmetry [6]. The C<sub>3v</sub> point group presents higher steric effect while C<sub>s</sub> has only one plane of symmetry. This, together with the fact that monoclinic P2<sub>1</sub>/c unit cell is smaller than the C2/c **causes that the inclusion of Lu<sup>3+</sup> ions strongly affects the crystal structure, despite their smaller ionic radius**. This way, the LGSO with GSO structure type allows low Lu doping. On the other hand, the steric effect is minor on a C2/c type structure and, furthermore, the Lu sites can be both occupied with larger size ions, such as Gd<sup>3+</sup> or Ce<sup>3+</sup> [6]. So, the symmetry degree of freedom should be quite higher than for GSO type. This implies that higher amounts of doping could be achieved. In fact, doping levels from x=0.17 up to 0.8 have been achieved for different researchers in the crystals with (Lu<sub>x</sub>Gd<sub>1-x</sub>)<sub>2</sub>SiO<sub>5</sub> compositions [5-14, 24].

For this reason, LGSO-3 and LGSO-5 have the  $C2/c$  crystal group although the Lu doping level is quite different.

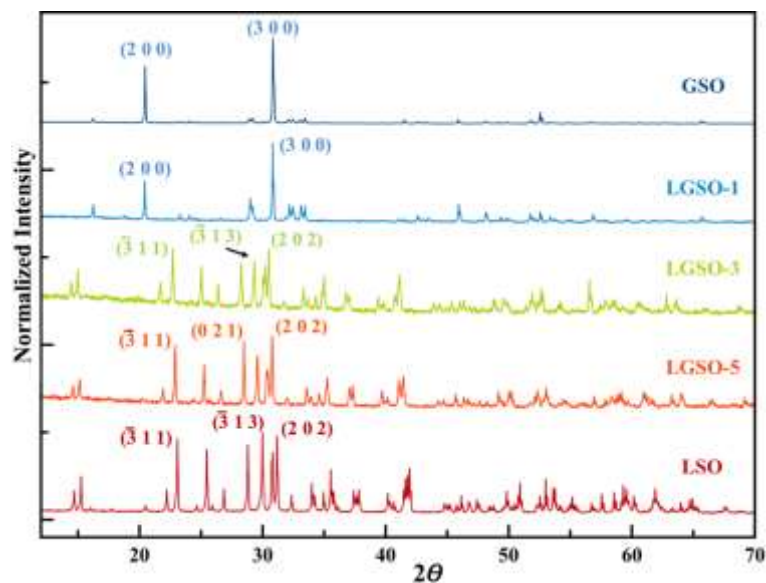


Fig. 3. Normalized XRD diffractograms of GSO (dark blue line, [16]), LGSO-1 (blue line), LGSO-3 (green line), LGSO-5 (orange line) and LSO (red line, [17]) sample powders. Most intense peaks are noticed.

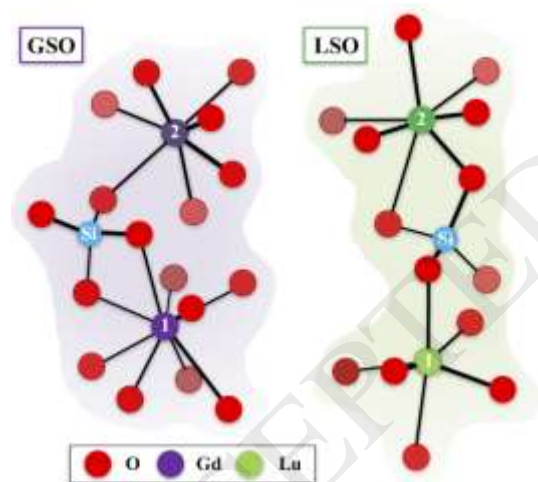


Fig. 4. **Structural** scheme of  $Gd_2SiO_5$  (GSO, left) and  $Lu_2SiO_5$  (LSO, right) **units** [5, 8, 9].

Coming back to LGSO fibres structure, LGSO-1 matched with the 01-080-9851 XRD card ICDD [23] that refers to the monoclinic  $P2_1/c$   $(Lu_{0.15}Gd_{0.85})_2SiO_5$  single crystal produced by Glowacki et al. [10] through CZ method. It must be noted that, despite the exhaustive milling, the powders diffractogram

maintains some degree of preferential orientations promoting defined reflections of the crystallographic planes more intense than others as was observed by Rey-García et al. [16] for the GSO grown by LFZ. This fact let us to confirm the high crystallinity degree of the fibres produced. The most intense peaks, located at 20.4° and 30.8°, would correspond to (2 0 0) and (3 0 0) planes respectively, extrapolating from the 01-080-9851 XRD card that referred the monoclinic  $(\text{Lu}_{0.15}\text{Gd}_{0.85})_2\text{SiO}_5$  oxyorthosilicate.

The diffractograms of LGSO-3 and LGSO-5 are isostructural with the 00-061-0488 and the 00-061-0369 XRD cards from the JCDD PDF4+ (Joint Committee for Diffraction Data, 2016) bibliographic data [23], corresponding to single crystal oxyorthosilicates with  $\text{Lu}_{0.36}\text{Gd}_{1.64}\text{SiO}_5$  and  $\text{Lu}_{1.33}\text{Gd}_{0.67}\text{SiO}_5$  compositions, respectively. Although a multitude of peaks can be observed in Fig. 3, the most intense ones are placed around similar degree values in both LGSO and un-doped LSO when extrapolated from the equivalent XRD cards. The peaks placed at 30.5° and 30.8° correspond to (2 0 2) plane and are the most intense in LGSO samples, while for the LSO the most intense is placed at 29.9° corresponding to  $(\bar{3} 1 3)$  plane [17]. Furthermore, the signals associated to  $(\bar{3} 1 1)$  plane are included between the three most intense peaks of each phase (Table III). All considered, it can be concluded that the introduction of  $\text{Gd}^{3+}$  ions on  $\text{Lu}_2\text{O}_3\text{-SiO}_2$  system avoids the formation of binary-eutectic fibres [17] **since the melting temperature is decreased, hampering the  $\text{SiO}_2$  evaporation, unavoided for the  $\text{Lu}_2\text{O}_3$  cubic ( $\text{Ia}\bar{3}$ ) C-type structure high melting point [21]. Therefore, the LSO C2/c monoclinic structure remains inside a Lu/Gd ratio ranging from 0.2 ( $(\text{Lu}_x\text{Gd}_{1-x})_2\text{SiO}_5$ ,  $x > 0.17$ ) to 1 (LGSO-5).**

**Table III.** XRD most intense peaks, corresponding planes and intensity for LGSO-3, LGSO-5 and LSO [17].

LGSO-3			LGSO-5			LSO [17]		
Peak	Plane	Intensity (%)	Peak	Plane	Intensity (%)	Peak	Plane	Intensity
22.7°	$(\bar{3} 1 1)$	99.7	22.9°	$(\bar{3} 1 1)$	89.8	23.2°	$(\bar{3} 1 1)$	92.6
29.3°	$(\bar{3} 1 3)$	85.8	28.5°	(0 2 1)	93.7	29.9°	$(\bar{3} 1 3)$	100
30.5°	(2 0 2)	100	30.8°	(2 0 2)	100	31.2°	(2 0 2)	95.5

Fig. 5 shows the Raman spectra of the LGSO samples compared with reported GSO and LSO Raman spectra [16, 17, 25]. The spectra were normalized to the intense peak recorded around  $890\text{ cm}^{-1}$  that defines the type of oxyorthosilicate present [25]. **Several** number of peaks have been recorded from each as-grown fibre. This fact is typical of oxyorthosilicates, **since**  $24\text{A}_g+16\text{B}_g$  to  $24\text{A}_g+18\text{B}_g$

**vibration modes can be observed** for GSO and LSO undoped materials, respectively [25]. The first approach allows observing how LGSO-1 spectrum is **close** to pure GSO [16] as expected. However, as Lu amount is increased, the Raman spectrum features change to shapes closer to un-doped LSO [17], as observed in LGSO-5 sample. On the other hand, the LGSO-3 fibre presents an intermediate spectrum.

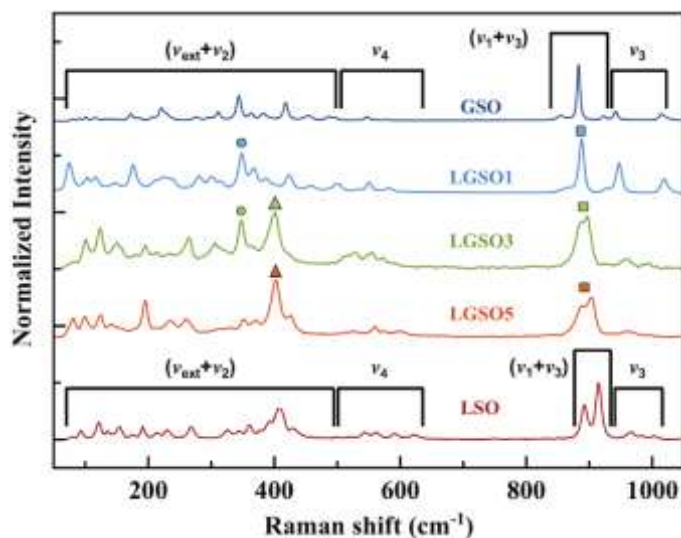


Fig. 5. Normalized Raman spectra of GSO (dark blue line, [16]), LGSO-1 (blue line), LGSO-3 (green line), LGSO-5 (orange line) and LSO (red line, [17]). GSO and LSO regions have been represented accordingly to Voron'ko et al. [25] notation. Bands discussed on Results section are denoted by circles, triangles and squares with the corresponding **color**.

By the way, the Raman spectra of the LGSO oxyorthosilicates developed by LFZ were studied according to the Voron'ko et al. notation [25]. Thus, the Raman spectrum of one defined oxyorthosilicate can be divided into four vibrational regions denoted by  $\nu_3$ ,  $\nu_1+\nu_3$ ,  $\nu_4$  and  $\nu_{\text{ext}}+\nu_2$ . Thus, the  $(\nu_1) - (\nu_4)$  modes correspond to the free  $[\text{SiO}_4]^{4-}$  internal vibrations in the reduced  $C_1$  symmetry of the monoclinic lattice and the  $(\nu_{\text{ext}})$  mode to the external oscillations related with the translation of the  $[\text{MO}_4]$ -complexes [25]. It must be mentioned that in compounds with tetrahedral oxyanions, there is a characteristic energy gap between 600 and 800  $\text{cm}^{-1}$  [10], which is also present here, and that usually separates internal bending and stretching vibrations of tetrahedral molecules. However, as Glowacki et al. [10] reported, the stretching vibrations of RE-O bonds can be recorded in the range of bending vibrations, hindering the clear identification of the peaks with their corresponding vibration modes.

Returning to the experimental results, the main feature takes place on the high frequency region ( $\nu_1+\nu_3$ ). A-type oxyorthosilicates present at around  $900\text{ cm}^{-1}$  the typical triplet of the GSO structures while the B-type oxyorthosilicates present a doublet degenerated band characteristic of LSO structures [16, 17, 25]. So, the LGSO-1 presents the triplet with the main peak at  $888\text{ cm}^{-1}$  (denoted in Fig. 5 by a blue square). The doublet associated to  $\nu_3$  antisymmetric stretching vibration of  $[\text{SiO}_4]^{4-}$  tetrahedron is also present at  $947$  and  $1019\text{ cm}^{-1}$ . In opposite, both LGSO-3 as LGSO-5 present two overlapped bands at  $887\text{-}897\text{ cm}^{-1}$  (green square) and  $890\text{-}903\text{ cm}^{-1}$  (orange square), respectively, which let us to consider them with LSO structure (Fig. 3, bottom, red spectrum) [8, 25]. To our knowledge, only Glowacki et al. [10] and Bynczyk et al. [8] reported Raman spectra for LGSO materials with GSO and LSO type structure, respectively. In the first case, the LGSO-1 composition is **close** to  $(\text{Lu}_{0.13}\text{Gd}_{0.865}\text{Sm}_{0.005})_2\text{SiO}_5$ , even considering that due to their small amount and size, the presence of Sm ions does not produce significant changes on the Raman spectrum [10]. So, Table IV summarizes the most significant recorded peaks, identified by software treatment of the data measured for LGSO fibres grown by LFZ. In the case of the LGSO-1 fibres, the Raman spectrum is compared to that obtained by Glowacki et al. [10] for their closer compositions, as mentioned above.

Focusing on the ( $\nu_{\text{ext}}+\nu_2$ ) region, which includes modes associated to rare earth ions (T(RE)), translatory and libratory modes associated to  $\text{SiO}_4$  anions (T, L( $\text{SiO}_4$ )), RE-O stretching vibrations ( $\nu(\text{RE})$ ), and bending vibrations of isolated  $\text{SiO}_4$  tetrahedrons ( $\nu_2(\text{SiO}_4)$ ) [10], there are two clear differences. The band placed at  $348\text{ cm}^{-1}$  for LGSO-1 (blue circle) and at  $349\text{ cm}^{-1}$  LGSO-3 (green circle) samples, was also reported for GSO by Voron'ko et al. [25] at  $344\text{ cm}^{-1}$  with  $A_g$  symmetry. Indeed, its intensity is much lower in the LGSO-5 fibres. Furthermore, the main characteristic band **is placed at  $403\text{ cm}^{-1}$  for LGSO-5 (orange triangle), at  $401\text{ cm}^{-1}$  for LGSO-3 (green triangle) and at  $408\text{ cm}^{-1}$  for LSO [17].** In fact, Zheng et al. [26] assigned this band, formed by  $405$  and  $411\text{ cm}^{-1}$  peaks, with the  $\nu_2$  bending mode. **Likewise**, Voron'ko et al [25] reported them at  $407$  and  $413\text{ cm}^{-1}$  with  $B_g$  and  $A_g$  symmetry, respectively. Thus, these bands, denoted by  $\nu_2(\text{RE-O})$  [10], are due to vibration modes associated to Gd and Lu **ions**, explaining their variation between materials even when the structure is similar.

**Table IV.** Raman peaks identified on LGSO fibres by software treatment. LGSO-1 peaks are compared to those identified by Glowacki et al. [10] for  $(\text{Lu}_{0.15}\text{Gd}_{0.845}\text{Sm}_{0.005})_2\text{SiO}_5$  (noted by 15Lu85Gd) with the corresponding

assignment of vibrations type. Notations from Voron'ko et al. [25] are also indicated. Wavenumber values are in  $\text{cm}^{-1}$ .

LGSO-1	15Lu85Gd [10]		Assignment [10]	LGSO-3	LGSO-5	Region [25]
	A <sub>g</sub>	B <sub>g</sub>				
75	-	-		80	83	
103	-	-		101	98	
117		115				
		132		123	125	
146	144			150	142	
				155	155	
176	175			171		
186		179		182	179	
205		209		195		
213	217		T (RE)	214	195	
223	222		T, L (SiO <sub>4</sub> )		214	
230	235		v(RE-O)	235	235	
239			v <sub>2</sub> (SiO <sub>4</sub> )		259	v <sub>ext</sub> +v <sub>2</sub>
280		279		264	285	
301		299		307	293	
315	314			322	315	
					330	
348	347			349	351	
367		366		365	370	
388	384					
	394			401	403	
424	424				427	
443		450				
458	457					
	491					
500		500		516	513	
535		531	v <sub>4</sub> (SiO <sub>4</sub> )	529	524	
550	548			550		v <sub>4</sub>
565	555			555	560	
583		581		572	573	
					600	
857	856					
861		865				
888	886			887	890	v <sub>1</sub> +v <sub>3</sub>
				897	903	
928		926				
947	947		v <sub>1</sub> , v <sub>3</sub> (SiO <sub>4</sub> )	930	935	
				950	945	
1019	1018			962	957	v <sub>3</sub>
				985	962	
		1026		997	970	

The bands placed below  $300\text{ cm}^{-1}$  are due to lattice vibrations of cation motion, modes of rare earth species and bending vibrations of the  $\text{SiO}_4$  tetrahedra, complicating their assignment [10]. Anyway, two bands can be ascribed based on bibliography. One is the band placed at 264 and  $259\text{ cm}^{-1}$ , in LGSO-3 and LGSO-5 fibres, respectively, which was assigned by Zheng et al. [26] to  $\nu_2$  bending modes of  $\text{SiO}_4$  in LSO. On the other hand, a band associated to  $\nu_4$  bending modes [26] is present around  $180\text{ cm}^{-1}$  in all cases, being more intense on LGSO-5 ( $179\text{ cm}^{-1}$ ). Additionally, no vibrational bands of  $\text{SiO}_2$  phases [27] were identified in any sample.

Finally, the optical behavior of the samples in the UV-visible range was evaluated and quantified by transmission spectroscopy (Fig. 6). This study gave us a first approach about their potential uses in photonic applications. The maximum transmittance measured, with values from 50% up to 77% for the visible range (Fig. 6a), corresponds to LGSO-3 sample as expected from the naked eye inspection (Fig. 1). In contrast, both LGSO-1 as LGSO-5 fibres present lower transmittance due to a scattering effect probably linked to the presence of internal microcracks. Furthermore, the birefringent character of the LSO [28, 29] influences these **measurements**. However, the birefringence of biaxial materials as LGSO or LSO, is considered not significant with respect to light output properties of the crystal when, for example, are employed as scintillators [30]. Concerning the LGSO-3 fibres, this composition, compared to un-doped oxyorthosilicates, seems to be optimal to reduce stress significantly increasing transparency and should provide good host materials. It must be noted that fringes due to interference phenomena provoked noise signal that are clearly observed on LGSO-3 and LGSO-5 spectra (Fig. 6a). Table V compared transmission values at selected wavelengths for each fibre.

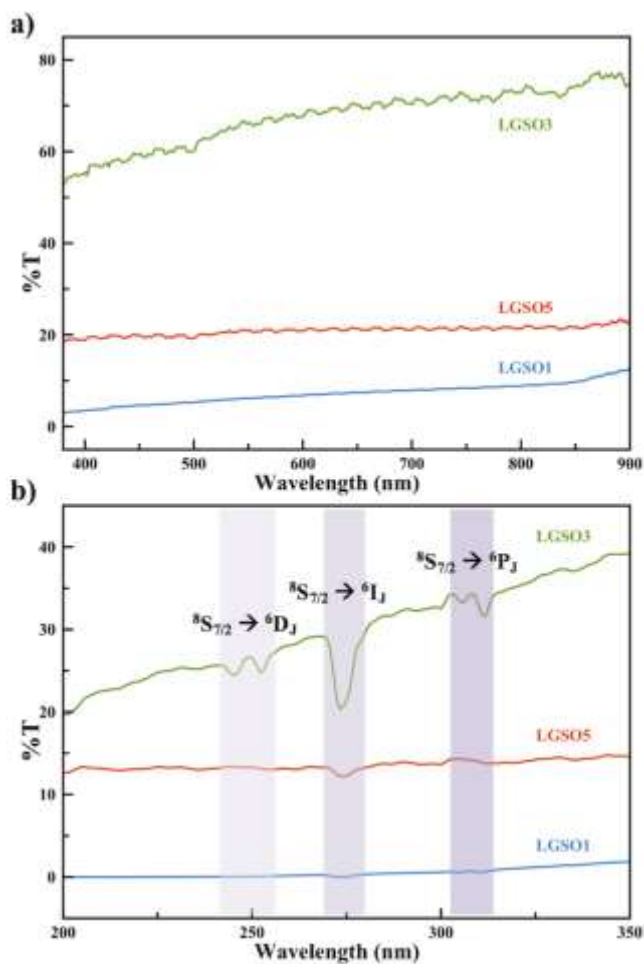


Fig. 6. Transmission spectra of LGSO-1 (blue line), LGSO-3 (green line) and LGSO-5 (orange line) at (a) visible (380-900 nm) and, (b) UV (200-350 nm) regions. Violet areas enclose the bands associated to intra  $4f^7$  transitions of  $Gd^{3+}$  ions from the  $^8S_{7/2}$  ground state [16].

**Table V.** LGSO fibres transmission values at selected wavelengths.

%T	Wavelength (nm)							
	200	275	400	500	600	700	800	900
Sample	200	275	400	500	600	700	800	900
LGSO-1	0.01	0.05	3.45	5.27	6.77	7.88	8.84	12.55
LGSO-3	19.6	21.36	55.27	60.29	67.56	70.27	72.83	75.23
LGSO-5	12.62	12.29	18.92	19.23	20.79	21.05	21.31	22.95



Focusing on UV range (200-350 nm) and **considering** the doping with optically active elements, a simple approach to photonic behavior was carried out centering our study on the areas corresponding to intra  $4d^7$  transitions of  $Gd^{3+}$  ions from the  $^8S_{7/2}$  ground state (Fig. 7b) [5, 16, 31, 32]. The main transfer band of the  $Gd^{3+}$  ions, that corresponds to  $^8S_{7/2} \rightarrow ^6I_1$  transition, is present in all samples at 275 nm [5, 16, 32]. However, its intensity is maximum for the LGSO-3 fibres while it is practically absent for LGSO-1. Furthermore, its intensity is higher than un-doped GSO developed by LFZ [16]. So, the LGSO-3 fibres should be presumably more efficient laser host materials than un-doped ones due to the energy transfer from  $Gd^{3+}$  ions to laser active dopant ions is higher. On the other hand, bands ascribed to  $^8S_{7/2} \rightarrow ^6D_J$  (245 and 252 nm) and  $^8S_{7/2} \rightarrow ^6P_J$  (306 and 311 nm) transitions [5, 16] were also identified at LGSO-3 spectrum. These correspond to those observed for GSO produced by LFZ [16]. Summarizing, the LGSO-3 fibres are good candidates to be doped with optically active ions and these should be optimal to be employed as host materials in photonic applications in agreement with the revised bibliography [10, 24].

## CONCLUSIONS

$(Lu_xGd_{1-x})_2SiO_5$  (LGSO;  $x=0.1, 0.3$  and  $0.5$ ) crystalline fibres have been grown in air by the Laser Floating Zone (LFZ) technique at pulling rates of 10 mm/h. The obtained fibres are homogeneous and the presence of macrostructural defects like cracks seems to be lower than observed in GSO fibres also obtained by LFZ. So, as expected, the effect of the  $Lu^{3+}$  substitution clearly reduces stress increasing plasticity. Additionally, it was observed that even with a Lu amount up to 0.5 mol.%, no formation of secondary phases like  $Lu_2O_3$  has been detected, in opposite to LSO fibres previously grown by LFZ.

The phase transition, namely from  $P2_1/c$  to  $C2/c$ , follows the rule established within the compositions tested. Indeed, the monoclinic  $P2_1/c$  (GSO-type) structure was obtained on LGSO-1 while materials with higher Lu content (LGSO-3 and LGSO-5) presented the  $C2/c$  structure (LSO-type). Concomitantly, the Raman spectra have also shown the structural transition from A-type (GSO type) oxyorthosilicates (LGSO-1) to B-type (LSO type) oxyorthosilicates (LGSO-5).

Finally, in the studied spectral range, the transmission studies of LGSO as-grown samples allow to conclude that the trivalent gadolinium ions are optically active presenting an intra-ionic absorption

due to the energy transfer band ascribed to  $^8S_{7/2} \rightarrow ^6I_1$  transition. This band is highest for LGSO-3 samples, which are the most transparent and present lower cracks. Thus, these facts let us to consider the LGSO-3 fibres **as host material** for future uses on photonic applications, i.e. as laser host materials.

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**Table I.** LFZ experimental conditions and comparison to pure GSO [16] and LSO [17] also grown in descendent direction on air by LFZ.

Sample	Nominal formula	Diameter (mm)	Pulling rate (mm/h)	Power (W)	Ref.
GSO	Gd <sub>2</sub> SiO <sub>5</sub>	1.5	10	72	<b>16</b>
LGSO-1	(Lu <sub>0.1</sub> Gd <sub>0.9</sub> ) <sub>2</sub> SiO <sub>5</sub>	1.5	10	67	-
LGSO-3	(Lu <sub>0.3</sub> Gd <sub>0.7</sub> ) <sub>2</sub> SiO <sub>5</sub>	1.5	10	58	-
LGSO-5	(Lu <sub>0.5</sub> Gd <sub>0.5</sub> ) <sub>2</sub> SiO <sub>5</sub>	1.5	10	64	-
LSO	Lu <sub>2</sub> SiO <sub>5</sub>	2	10	92	<b>17</b>

**Table II.** Initial and calculated compositions determined by EDS analysis.

Sample	Initial composition	EDS calculated composition
LGSO-1	$(\text{Lu}_{0.1}\text{Gd}_{0.9})_2\text{SiO}_5$	$(\text{Lu}_{0.12}\text{Gd}_{0.88})_2\text{SiO}_5$
LGSO-3	$(\text{Lu}_{0.3}\text{Gd}_{0.7})_2\text{SiO}_5$	$(\text{Lu}_{0.31}\text{Gd}_{0.69})_2\text{SiO}_5$
LGSO-5	$(\text{Lu}_{0.5}\text{Gd}_{0.5})_2\text{SiO}_5$	$(\text{Lu}_{0.53}\text{Gd}_{0.47})_2\text{SiO}_5$

**Table III.** XRD most intense peaks, corresponding planes and intensity for LGSO-3, LGSO-5 and LSO [17].

LGSO-3			LGSO-5			LSO [17]		
Peak	Plane	Intensity (%)	Peak	Plane	Intensity (%)	Peak	Plane	Intensity
22.7°	$(\bar{3} 1 1)$	99.7	22.9°	$(\bar{3} 1 1)$	89.8	23.2°	$(\bar{3} 1 1)$	92.6
29.3°	$(\bar{3} 1 3)$	85.8	28.5°	$(0 2 1)$	93.7	29.9°	$(\bar{3} 1 3)$	100
30.5°	$(2 0 2)$	100	30.8°	$(2 0 2)$	100	31.2°	$(2 0 2)$	95.5

**Table IV.** Raman peaks identified on LGSO fibres by software treatment. LGSO-1 peaks are compared to those identified by Glowacki et al. [10] for  $(\text{Lu}_{0.15}\text{Gd}_{0.845}\text{Sm}_{0.005})_2\text{SiO}_5$  (noted by 15Lu85Gd) with the corresponding assignment of vibrations type. Notations from Voron'ko et al. [25] are also indicated. Wavenumber values are in  $\text{cm}^{-1}$ .

LGSO-1	15Lu85Gd [10]		Assignment [10]	LGSO-3	LGSO-5	Region [25]
	A <sub>g</sub>	B <sub>g</sub>				
75	-	-	T (RE) T, L (SiO <sub>4</sub> ) v(RE-O) v <sub>2</sub> (SiO <sub>4</sub> )	80	83	v <sub>ext</sub> +v <sub>2</sub>
103	-	-		101	98	
117		115		123	125	
		132		150	142	
146	144			155	155	
				171		
176	175			182	179	
186		179		195		
205		209		214	195	
213	217			235	235	
223	222				259	
230	235			264	285	
239				307	293	
280		279		322	315	
301		299			330	
315	314			349	351	
				365	370	
348	347			401	403	
367		366			427	
388	384					
	394					
424	424					
443		450				
458	457					
	491					
500		500		516	513	
535		531		529	524	
550	548			550		
565	555			555	560	
583		581		572	573	
					600	
857	856					
861		865				
888	886			887	890	
				897	903	
928		926				
947	947			930	935	
		v <sub>1</sub> , v <sub>3</sub> (SiO <sub>4</sub> )		950	945	
1019	1018			962	957	
				985	962	
		1026		997	970	



**Table V.** LGSO fibres transmission values at selected wavelengths.

%T	Wavelength (nm)							
Sample	200	275	400	500	600	700	800	900
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LGSO-3	19.6	21.36	55.27	60.29	67.56	70.27	72.83	75.23
LGSO-5	12.62	12.29	18.92	19.23	20.79	21.05	21.31	22.95