Evaluation of ketoprofen toxicity in two freshwater species: Effects on biochemical, physiological and population endpoints

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### 23 Abstract

Among the most used non-steroidal anti-inflammatory drugs (NSAIDs), ketoprofen 24 25 (KTF) assumes an important position. Nevertheless, its ecotoxicological effects are in 26 non-target organisms poorly characterized, despite its use and frequency of occurrence in aquatic matrices. Thus, the aim of this study was to evaluate the possible 27 toxicological effects of KTF contamination, in two freshwater species, Lemna minor 28 and Daphnia magna, by measuring biochemical, physiological and population 29 parameters. To attain this objective, both species were exposed to KTF at the same 30 concentrations (0, 0.24, 1.2, 6 and 30 µg/L). L. minor plants were exposed during 4 d to 31 these levels of KTF, and the enzymatic activity (catalase (CAT), glutathione S-32 transferases (GSTs) and carbonic anhydrase (CA)), and pigments content (chlorophyll a, 33 b and total and carotenoids) were analyzed to evaluate the toxicity of this drug. D. 34 magna was acutely and chronically exposed to KTF, and enzymatic activities (CAT, 35 GSTs and cyclooxygenase (COX)), the feeding rates, and reproduction traits were 36 37 assessed. In L. minor, KTF provoked alterations in all enzyme activities, however, it was not capable of causing any alteration in any pigment levels. On the other hand, KTF 38 also provoked alterations in all enzymatic activities in D. magna, but did not affect 39 feeding rates and life-history parameters. In conclusion, exposure to KTF, provoked 40 biochemical alterations in both species. However, these alterations were not reflected 41 into deleterious effects on physiological and populational traits of L. minor and D. 42 magna. 43

44 Capsule: KTF demonstrates species-specific toxicity to different aquatic organisms
45 affecting biochemical levels but not impairing physiological and population levels.

46 Keywords: Non-steroidal anti-inflammatory drugs; *Lemna minor*; *Daphnia magna*;
47 Photosynthetic pigments; Biochemical markers; Reproduction.

Thousands of tons of pharmaceutical substances are used yearly to prevent or treat 49 illnesses (Kosjek et al., 2005; Lapworth et al., 2012). Drugs have been recognized as a 50 51 large class of chemical contaminants which may originate from human and aquaculture usages, as direct results of excretions of metabolites and residues after metabolism 52 (Eslami et al., 2015), among other sources. The presence and potential effects of these 53 54 chemicals has been gaining attention due to the increase of their discharges, leading to an augmented frequency of their detection in the environment (Kosjek et al., 2005; 55 Rzymski et al., 2017). 56

57 Non-steroidal anti-inflammatory drugs (NSAIDs) are among the most frequently prescribed drugs in modern medicine (Meek et al., 2010). It was estimated that the 58 worldwide NSAIDs production is of several kilotons each year (Cleuvers 2004; 59 Freches, 2017). These drugs are commonly used in the treatment of symptoms such as 60 inflammation, pain, and fever (Simon, 2013; Al-Khateeb et al., 2017). This usage is due 61 62 to their pharmacological activity, that occurs through cyclooxygenase (COX) inhibition, enzyme responsible for the production off prostaglandins, which are chemical mediators 63 involved in inflammatory processes (Hernando et al. 2006). Beyond its use for the 64 65 general purposes that NSAIDS are used for, KTF may also be used to treat rheumatoid arthritis and osteoarthritis, and to relieve muscle and joint pain (Prášková et al., 2011). 66 Among NSAIDs that are nowadays in use for human therapeutics, it is possible to 67 identify ketoprofen (KTF), which is the third most used NSAID; data from the literature 68 show that, in Croatia, and between 2007 and 2013, the numbers of KTF consumption 69 70 were just behind values for diclofenac and ibuprofen (Krnic et al., 2015). In 2015, in England, a total of about 25,400 prescriptions for KTF were accounted for (PACT 71 2016), while in 2017 this value reached 147,971, in the USA (ClinCalc, 2020). 72

KTF, in humans, is extensively and rapidly metabolized by the liver, mainly via conjugation with glucuronic acid (Gierse et al., 1999) and approximately 80% of the administered dose is excreted in the urine in a 24-hour period after the administration (Prášková et al., 2011); however, there is no information about these parameters for aquatic organisms, which is now critically required for the assessment of its putative toxicological environmental assessment.

NSAIDs are one of the most frequently detected pharmaceutical compounds in the 79 aquatic environment, and their environmental distribution is widespread (Gentili 2007; 80 aus der Beek et al., 2016, including KTF. These drugs, in general, are not efficiently 81 eliminated in sewage treatment plants, being released to surface waters (Martinez-Sena 82 et al., 2016). Thus, KTF is reported in aquatic environment up to 0.3 µg/L and up to 83 0.87 µg/L in Catalonia (Spain) rivers and effluents respectively (Farré et al., 2001), 84 around 0.2 µg/L in German municipal sewage treatment plant (Ternes, 1998), 0.18 µg/L 85 (Tixier et al, 2003) in Switzerland waste water treatment plant and 5.7 µg/L in effluents 86 of Canadian sewage treatment plants (Metcalfe et al., 2003), besides that, KTF was 87 recorded up to 2 µg/L in Finland influents (Lindqvist et al., 2005). 88

Considering the large number of NSAIDs, in general, their environmental fates and 89 90 effects are still poorly understood, despite the already published data that evidence their persistent nature, and bioaccumulation potential in aquatic organisms (Geng et al., 91 2018). This is a pertinent gap in the knowledge of the environmental effects of drugs, 92 since NSAIDs were designed to be biological active in a specific group of organisms 93 (Kosjek et al., 2005; Ghlichloo and Gerriets, 2020). Mammalians, for instance, have 94 both COX isoforms (COX-1 and COX-2, mainly), enzymes responsible for the 95 regulation and onset of the inflammatory process (Heckmann et al., 2008a; 96 Wongrakpanich et al., 2018). However, one COX isoform, at least, is generally present 97

in invertebrates and lower vertebrates (Rowley et al., 2005) and, consequently, these 98 organisms may also be affected by NSAIDs too. Consequently, these pharmaceuticals 99 may be responsible for a large number of toxicological effects in non-targets organisms, 100 including macrophyte species (Alkimin et al., 2019a), mollusks (Almeida and Nunes, 101 102 2019; Piedade et al., 2020), polychaetes (Gomes et al., 2019) and fish (Nogueira et al., 2019). Toxicological effects of KTF have been mostly limited to a few studies, that 103 measured its toxic effects on embryo-larval development and biochemical alterations, 104 105 including the oxidative stress response, assessed in both larvae and adults of the fish species Danio rerio (Diniz et al., 2015; Rangasamy et al., 2018). Toxic effects of KTF 106 were also determined in the freshwater fish Cyprinus carpio (Prášková et al., 2013), 107 also focusing in embryo-larval development. In addition, KTF toxicity was analyzed in 108 the microcrustacean *Daphnia magna*, and obtained data evidenced its capacity to alter 109 110 the swimming behavior and physiological endpoints (heart rate, thoracic limb activity, and mandible movement) in this organism (Bownik et al., 2020) 111

Besides their pharmacological activity, which occurs through COX inhibition, some 112 NSAIDs have the ability to interact with biota, thereby affecting other biological 113 parameters. In humans, NSAIDs are capable of inhibiting the carbonic anhydrase (CA) 114 activity (Knudsen et al., 2004), an enzyme responsible for carbon dioxide hydration 115 (Lindskog, 1997). NSAIDs also induce the generation of reactive oxygen species (ROS) 116 (Galati et al., 2002; Adachi et al., 2007; Ito et al., 2016) that can lead to oxidative stress. 117 118 The effects of NSAIDs are not limited to the here-described toxicity, since they may also directly compromise the eicosanoids biosynthesis, an important molecule in 119 120 reproductive traits of crustaceans (Heckmann et al., 2008a), and provoke physiological 121 alterations on macrophyte species (Wrede, 2015). Thus, it is possible to assume that 122 more studies about these effects associated to NSAIDs exposure, namely KTF, are

123 urgently required, in a comprehensive range of organisms. Besides the necessity to study the effects of KTF on a comprehensive range of distinct test organisms, it is 124 equally necessary to determine its effects on a set of toxicological complimentary 125 endpoints, from different levels of organization, including at the biochemical, 126 physiological and population levels. This will allow creating a body of evidence about 127 KTF toxicity to aquatic organisms, which is nowadays mostly inexistent. In addition, 128 these studies were performed by testing much higher concentrations than those found in 129 130 the aquatic environment, considering the already mentioned studies reporting the environmental presence and levels of this drug. To fully characterize the 131 ecotoxicological potential of this drug, it is necessary to determine its toxicity at 132 environmentally realistic concentrations, in aquatic organisms. Thus, the aim of this 133 study was to evaluate the toxicological effects of KTF at realistic environmental 134 135 concentrations and worst scenario case scenarios, in two freshwater species, namely a macrophyte species (Lemna minor) and a microcrustacean (Daphnia magna), by 136 137 measuring biochemical, physiological and populational parameters.

138

### 139 Material and methods

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### 141 Chemicals

142 Ketoprofen (CAS number: 22071-15-4) was purchased from Sigma-Aldrich (Belgium) 143 with  $\geq$  98% of analytical purity; all other chemicals used in this study (buffers, protein 144 determinations) had analytical purity and were purchased from Sigma Aldrich and Bio-145 rad® laboratories.

### 147 Organisms culture and maintenance

All organisms (*Lemna minor* and *Daphnia magna*) were cultured and maintained at the
Department of Biology at Aveiro University, as described by Alkimin et al. (2019b) for *L. minor*, and by Daniel et al. (2019) for *D. magna*.

151

# 152 Exposures of test organisms

All experiments performed in this work involved exposing both species to the same range of concentrations, based on already reported environmental levels (up to 5.7  $\mu$ g/L) (Farré et al., 2001; Metcalfe et al., 2003; Lindqvist et al., 2005). The nominal exposure concentrations were: 0, 0.24, 1.2, 6 and 30  $\mu$ g/L; the control treatments involved exposing organisms only to the culture medium of each species. The KTF stock solutions were directly prepared in each specific culture medium by dissolving the pure compound.

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### 161 Macrophytes – Lemna minor

162 Plants of the macrophyte species L. minor were exposed in a 6-well plate, filled with 10 163 mL of medium and/or stock solution per well. The test started with macrophytes covering around 20% of the surface of each well, with 10 replicates per condition. The 164 165 test had 96 h of duration, a period that was chosen considering the assumptions described by Alkimin et al. (2019b). This study established that 96h is a duration long 166 enough to cause measurable alterations in Lemna species after being exposed to 167 xenobiotics, including to some pharmaceutical drugs. Exposures were conducted 168 according to OECD (2006) under controlled conditions (temperature  $23 \pm 1$  °C; 169

170 continuous light exposure; light intensity, ~84  $\mu$ mol photons m<sup>-2</sup> s<sup>-1</sup>) in a climate 171 chamber (Binder). After the exposure period, *Lemna* fronds from each treatment were 172 collected, and this biomass was divided and placed in Eppendorfs microtubes, and 173 stored at -80 °C for the analyzes - quantification of pigments (chlorophyll a, b and total 174 and carotenoids) and determination of biochemical endpoints (catalase, CAT; 175 glutathione-S-transferases, GSTs; and carbonic anhydrase, CA, activities).

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# 177 Microcrustacean – Daphnia magna

Three different experiments were performed with D. magna. In the first one, juveniles 178 (5 d old) were exposed to KTF during 48 h in 300 mL glass flasks filled with 200 mL of 179 KTF or medium solution. For each concentration, 5 replicates with 12 organisms each 180 181 were adopted. This experiment was repeated thrice, each one for obtaining biological samples to quantify a specific enzyme (biochemical markers). At the end of the 182 exposure periods, animals were collected with a plastic pipette to Eppendorf microtubes 183 184 and stored at -80 °C for ulterior analysis (determination of CAT, GSTs, and COX activities). Feeding rate experiments were conducted in test-chambers consisted of 185 50 mL glass flasks filled with test solution, where five 5 d old daphnids were positioned 186 (five replicates per treatment). Briefly, this test was used to measure the filtration 187 capacity on algal cells (Raphidocelis subcapitata) by test organisms, when exposed to 188 189 the drug. The animals were exposed for 24 h, in the dark, to all experimental conditions, with  $5 \times 10^5$  cells/mL of the algae *R*. subcapitata in the test-chambers. Afterwards, 190 animals were transferred to a new test-chamber (with clean media), also in the dark, for 191 192 4 h, with the same algae density (post-exposure period), to evaluate the recovery from exposure to toxicants. Blank controls (media with no daphnids) were included in the 193 experimental design in both cases. At the end of each period, the algae cell densities 194

195 were spectrophotometrically measured (Thermo Scientific Multiskan (ScanIt Software 2.4.4)) at 440 nm. Finally, the feeding rate was calculated converting the obtained 196 values to proportional algae consumption (%) relative to control. Finally, a chronic 197 reproduction test was conducted, based on the OECD guideline 211 (OECD 2012) with 198 the exposure period modified according to Ribeiro et al. (2011) and Alkimin et al. 199 (2020). The test duration was approximately 16 days and/or the third brood. The 200 experiment was performed in 50 mL glass flasks, and 10 replicates with < 24 h old 201 202 neonates were used for each concentration. Medium test was totally renewed every other day, and the animals were maintained in the same culture conditions. The 203 parameters day of first brood, number of neonates from first brood, and total number of 204 neonates were the evaluated endpoints. 205

206

# 207 Quantification of physiological and biochemical parameters

# 208 Pigments analysis (chlorophylls and carotenoids)

Total, a, and b chlorophylls (TChl; Chl a; Chl b), and carotenoids (Car) amounts were 209 210 determined spectrophotometrically, according to the method described by Hiscox and 211 Israelstam (1979). Pigments were extracted from the previously exposed fronds of L. 212 minor (about 10 mg per replicate - fresh weight, FW) in 1.8 mL of dimethyl sulphoxide 213 (DMSO). The extract was placed in water at 65 °C during 30 min and allowed to cool in 214 the dark and at room temperature. The obtained supernatants were used to quantify the amounts of chlorophylls (a, b, and total), and carotenoids levels, by spectrophotometry, 215 by measuring the absorbance of the extracts, at wavelengths of 470, 645, 646, and 663. 216 The calculation of the pigments followed the equations proposed by Arnon (1949) and 217 demonstrated by Hiscox and Israelstam (1979) to be suitable if the extraction was 218

undertaken with DMSO, to calculate the amounts of Chl (a, b, and total), and by
Lichtenthaler (1987) to quantify the Car amount. In addition, two ratios were
determined: [Chl a] / [Chl b]; and [TChl] / [Car].

222

# 223 Tissue homogenization – biochemical markers

For CAT and GSTs determinations, samples were macerated with mortar and pestle (L. 224 minor) or sonicated (D. magna - Branson sonicator, model 250) in ice-cold phosphate 225 buffer (50 mM, pH 7.0, with 0.1% Triton X-100). Each homogenate sample was 226 composed by ~ eight fronds of L. minor, or 12 individuals of D. magna, and 1.2 mL 227 phosphate buffer. Homogenized samples were centrifuged at 15,000 g at 4°C for 10 228 min. For carbonic anhydrase activity (CA) determination, samples of L. minor were 229 230 macerated in 1 mL ice-cold Tris-sulfate 25 mM (pH 7.5) with 25 mM of sodium sulfate. Homogenized samples were centrifuged at 10,000 g at 4 °C for 40 min. Cyclooxygenase 231 determination (D. magna) was performed using 0.8 mL of 0.1 M Tris-HCl buffer (pH 232 233 7.8) with 1 mM EDTA to sonicate animals, followed by a cycle of centrifugation at 10,000 g at 4 °C for 15 min. After the centrifugation process, supernatants were divided 234 into aliquots, which were used for the different enzymatic determinations, as described 235 below. 236

237

### 238 Catalase activity determination

Catalase activity was determined in a 96-well microplate, by the procedure described by Aebi (1984), based on the degradation rate of the substrate  $H_2O_2$ , monitored at 240 nm for 5 min. The results were expressed by considering that one unit of CAT activity equals the number of moles of  $H_2O_2$  degraded per minute, per milligram of protein.

### 243 Glutathione-S-transferases activity determination

The procedure to determine GSTs activity monitored the increment of absorbance at 340 nm of a thioether resulting from the catalysis of the substrate 1-chloro-2,4dinitrobenzene (CDNB) with glutathione by GSTs, as described by Habig et al. (1974), adapted to a 96 well microplate. Results were expressed as millimoles of thioether produced per minute, per milligram of protein.

249

### 250 Carbonic anhydrase activity determination

To determine CA activity, the method by Verpoorte et al. (1967) was adapted to 96-well microplates. This methodology is based on the hydrolysis of the substrate p-nitrophenol acetate (pNPAc), catalyzed by CA. This reaction was then monitored for 5 min at a wavelength of 400 nm. One unit of CA is defined as the amount of enzyme needed to hydrolyze one µmol of pNPAc per minute, and the data were expressed in U/milligram of protein.

257

# 258 Cyclooxygenase activity determination

This method is based on the ability of cyclooxygenase (COX) to convert arachidonic acid into a hydroperoxy endoperoxide (Prostaglandin G2; PGG2). This is reduced in the presence of N, N, N', N'-tetramethyl-p-phenylenediamine (TMPD) to its alcohol (Prostaglandin H2; PGH2); then, the oxidation of TMPD was monitored during 5 min at 590 nm (Petrovic and Murray, 2010).

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### 266 **Protein determination**

Protein quantification was performed at 595 nm using the Bradford method (Bradford 1976), adapted to microplate, with bovine  $\gamma$ -globulin as standard (1mg/ml), in order to express enzymatic activities per mg of protein of the analyzed samples. Finally, all spectrophotometric readings were performed in a microplate reader Thermo Scientific Multiskan (ScanIt Software 2.4.4).

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# 273 Statistical analysis

Data for all parameters were tested for homogeneity and normality, and a one-way analysis of variance (ANOVA) was performed, followed by a post-hoc Dunnett's test (or the equivalent non-parametric test, according to the data). Statistical analysis was performed with SigmaPlot v.14.0. The adopted level of significance was 0.05.

278

### 279 **Results**

The lowest concentration (0.24 µg/L) of KTF provoked a decrease in GSTs activity in 280 281 L. minor; on the contrary, this enzyme's activity was increased in plants exposed to all other concentrations (1.2, 6, and 30  $\mu$ g/L) (F<sub>(4, 28)</sub> = 57.177, p <0.001; Figure 1A). With 282 the exception of the highest concentration (30  $\mu$ g/L), KTF increased CAT activity (F <sub>(4,</sub> 283 284 <sub>32)</sub> = 23.157, p<0.001; Figure 1B). KTF caused a decrease of CA activity in plants exposed to levels of 0.24 and 1.2  $\mu$ g/L (F <sub>(4, 46)</sub> = 10.055, p<0.001; Figure 1C) and no 285 effect was observed in plants exposed to the two highest concentrations (6, and 30 286  $\mu g/L$ ). 287

Photosynthetic pigments (Chl a, b and TChl) measured in *L. minor* did not suffer any alteration after KTF exposure ( $F_{(4, 47)} = 0.922$ , p = 0.460;  $F_{(4, 47)} = 1.426$ , p = 0.242 and  $F_{(4, 47)} = 1.003$ , p = 0.416, respectively). No alterations were also detected for Car content ( $F_{(4, 47)} = 1.139$ , p = 0.351,) or for both determined ratios (Chl a/b – H<sub>4df</sub> = 4.255, p = 0.373; TChl/Car –  $F_{(4, 47)} = 2.022$ , p = 0.108) (Figure 2A – F).

- In *D. magna*, KTF caused an increase in CAT activity ( $F_{(4,20)}=31.597$ ; p<0.001) in a concentration related manner (Figure 3A); GSTs activity also presented an increase, but only statistically different in organisms exposed to concentrations of 1.2 and 6 µg/L ( $F_{(4,22)}=5.048$ ; p=0.007)(Figure 3B). On the other hand, COX activity was reduced after KTF exposure ( $F_{(4,18)}=4.695$ ; p=0.013) for organisms exposed to all tested concentrations, in a dose-response manner (Figure 3C).
- Ketoprofen did not affect *D. magna* feeding rates during exposure ( $F_{(4,24)}$ = 3.515; p = 0.025 ) or during recovery period ( $F_{(4,24)}$ = 1.636 p=0.0204; Figure 4).

Ketoprofen did not affect reproductive endpoints in *D. magna*, which included day of first brood ( $H_{4df}$  =4.091; p=0.394; Figure 5A), number of neonates from first brood ( $H_{4df}$ =2.803; p=0.591; Figure 5B), and total number of neonates ( $F_{(4, 44)}$ =0.938; p=0.452; Figure 5C).

305

### 306 **Discussion**

307 Lemna minor

NSAIDs, including diclofenac and paracetamol, are capable to induce ROS production in *L. minor*, as already demonstrated by Kummerová et al. (2016). This is highly important in our context, considering that reactive oxygen species (ROS), and

particularly hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>), are generated during the normal plant 311 metabolism and are involved in a diversity of signaling cascades and reactions 312 necessary for the physiological regulation of diverse aspects, such as plant growth 313 (Cheeseman, 2007). Exposure to anthropogenic xenobiotics, such as the referred drugs, 314 315 may enhance this production, leading to toxic effects in plants. A similar finding was made clear in this study, since CAT activity was increased following KTF exposure. 316 Increased CAT activity must be understood as the activation of enzymatic detoxification 317 318 mechanisms that to minimize cellular levels of superoxide radicals  $(O_2^-)$  and  $H_2O_2$ . This is the main role of antioxidant enzymes such as CAT (Sairam and Tyagi, 2004). It is 319 thus possible to hypothesize that low levels of KTF were causative of the triggering of 320 an antioxidant response. Conversely, the activity of CAT measured in plants exposed to 321 the highest KTF concentration was closely similar to the control values; this can be 322 323 assumed as a possible plant resilience mechanism, since macrophytes are known for their ability to recover along time (Wang, 1990), when kept in a contaminated 324 325 environment by different contaminants, such as metals (Jaramillo et al., 2019) and 326 pharmaceutical drugs (Alkimin et al., 2019a). The normalization of CAT means that plants were able to activate alternative detoxification and antioxidant mechanisms, such 327 as ascorbate peroxidase and superoxide dismutase, another two enzymes very important 328 329 for the antioxidant defense system of plants (Mittler, 2002). For example, the different affinities of ascorbate peroxidase (APX; µm range) and CAT (mm range) for H<sub>2</sub>O<sub>2</sub> 330 indicated that they belong to different classes of ROS-scavenging enzymes, namely, 331 H<sub>2</sub>O<sub>2</sub> (Mittler, 2002) This means that the fine modulation of ROS signaling might 332 occurs through APX, whereas CAT might be responsible for the removal of excess ROS 333 334 during stress (Mittler, 2002).

335 The alterations of CAT activity may be linked to the observed changes in GSTs activity. Plants exposed to the lowest concentration of KTF had a decrease in GSTs activity. It is 336 possible to suggest that the overproduction of ROS by KTF exposure (signaled by the 337 enhancement of CAT activity) may have caused direct damages in cellular molecules. 338 This a common feature of ROS, and their damaging action may include deleterious 339 alterations of DNA, lipids, and proteins; this often causes a decrease in enzymatic 340 activity by denaturation (Schieber and Chandel, 2014), an effect that may be held 341 342 accountable for the here-observed decrease of GSTs activity. On the other hand, plants exposed to the other tested concentrations were capable of increasing their GSTs 343 activity. This pattern is indicative of distinct effects, related to the increase of 344 biotransformation capacity of the organism to cope with the excess of this xenobiotic. 345 Since, this isoenzyme group is part of phase II metabolic mechanisms that is responsible 346 347 for the detoxification of exogenous compounds; conjugating glutathione (GSH) with compounds containing an electrophilic center to modify the substrate into a more water 348 349 soluble, less toxic complex (Habig et al., 1974). In addition, GSTs had also a role in 350 resisting pro-oxidative effects, being useful in enhancing plant survival on toxic sites (Cummins et al., 2011). This effect occurs since GSTs contribute for the efficacy of the 351 antioxidant defense system, which does not directly relate to their role in xenobiotic 352 353 detoxification (Moons, 2005).

Macrophytes (such as *L. minor*) are primary producers in the aquatic environment, consequently being at the basis of the food web of freshwater systems; thus, adverse effects that may occur in these plants can also deleteriously affect the entire aquatic food web. So, if the *L. minor* antioxidant stress system was not able to efficiently reduce the adverse effect of xenobiotics upon exposure, deleterious effects might occur in this species, from cell death to other unpredictable consequences (Demidchik, 2015; Xie et

al., 2019). Despite the general absence of data for plant organisms, it is possible to consider the assumptions made by Monaghan et al. (2009), when referring that life history trade offs in animals are likely to be associated to oxidative stress and to the antioxidant response, which occur at the most fundamental metabolic activities of living organisms. Such traits are likely to impact not only the individual, but to escalate at higher levels of organization, with unforeseen outcomes. So, changes in plants of *L. minor* may indeed correspond to alterations at the ecosystem level.

Carbonic anhydrase is a zinc-containing enzyme that catalyzes the rapid conversion of 367 CO<sub>2</sub> over water in a proton and bicarbonate ion (HCO<sub>3</sub><sup>-</sup> - reversible 368 hydration/dehydration reaction) (Coleman, 2000; Escudero-Almanza et al., 2012). In 369 plants, CA is important in many physiological functions that involve carboxylation or 370 decarboxylation reactions (Moroney et al., 2001) by increasing CO<sub>2</sub> concentrations in 371 372 the chloroplast to increase the carboxylation rate of the ribulose 1,5-disphosphate carboxylase (RuBisCO) enzyme. This chemical reaction is responsible for the 373 374 incorporation of CO<sub>2</sub> into carbohydrates during photosynthesis; however, it and can 375 only use C coming from CO<sub>2</sub>, instead of C from carbonic acid or bicarbonate (Escudero-Almanza et al., 2012). In addition, CA also actively participates in the 376 377 inorganic carbon transportation into actively photosynthesizing cells or away from 378 actively respiring cells (Henry 1996). Few studies reported the effects of pharmaceutical drugs in CA enzymatic activity of macrophytes. Previous data have shown that 379 380 acetazolamide (diuretic drug) and salicylic acid (NSAID), had the capacity to decrease CA activity, also in a macrophyte species, namely L. gibba (unpublished data). 381 382 However, the here obtained data showed a decrease of CA activity only in plants 383 exposed to the lowest concentrations of KTF (0.24 and 1.2 µg/L), which may indicate 384 hormesis. In general, a hormesis effect represents an adaptive response of organisms to

environmental or self-imposed challenges through which they improve its functionality
and/or tolerance to more aggressive challenges (Calabrese and Mattson, 2017). Such
challenging conditions, in this case, correspond to the higher concentrations of the
tested pharmaceutical. The exposed plants showed to be resilient, by decreasing their
CA activity. Hormesis effects were also observed in CAT activity, which only increased
in plants exposed to the lowest concentrations.

391 Generally, higher plants have three groups of pigments: carotenoids, chlorophylls, and 392 phycobilins; the most important pigments involved in photosynthesis are chlorophylls, which are prone to be targets for toxic alterations. Changes of the qualitative 393 composition or of quantitative content of pigments, are physiologically important 394 characteristics that indicate the general health condition of plants, informing also on the 395 function of photosynthetic apparatus, including the adaptive responses after being 396 397 subjected to stressful situations (Belous et al., 2018). Levels of the main photosynthetic pigments here analyzed (Chl a, b and total) did not suffer any alteration after KTF 398 exposure; in addition, the ratio between levels of chlorophylls a and b were not altered. 399 400 This ratio is a calculation that characterizes the photosynthetic apparatus operation, by indicating the potential photochemical activity of leaves. The absence of effects may 401 402 lead us to conclude that KTF does not compromise the normal photochemical potential 403 in L. minor, despite pervious indications that, in general, NDSAIDs may exert this effect. As far as we know, no data are available concerning this parameter, for 404 405 macrophyte species after being exposed to KTF. However, as mentioned, NSAIDs are 406 capable to cause alterations in chlorophyll content in plants. For example, diclofenac (100  $\mu$ g/L; 314.3 nM) and paracetamol (100  $\mu$ g/L; 661.5 nM) were capable to decrease 407 408 the contents of photosynthetic pigments in L. minor after 7 days of exposure 409 (Kummerová et al., 2016). Diclofenac (1 mg/L; 3.1 mM) was also responsible for the

decrease of Chl a and b content in chicory (Cichorium intybus), after 22 days of 410 exposure (Podio et al., 2020). Thus, we can hypothesize that environmental 411 concentrations of KTF are safe to Lemna species in relation to this specific endpoint, 412 being KTF a less toxic pharmaceutical drug among the NSAID class to this genera. We 413 may also suggest that the here studied period of exposure was not long enough to trigger 414 alterations in this endpoint after KTF exposure. This is an important set of results, since, 415 to the best of our knowledge, this is the first report about KTF toxicity evaluation in 416 417 Lemna species.

Carotenoids are well known free-radical scavengers, also playing an important role in 418 photoprotection of photosynthetic apparatus (Sytar et al., 2013). KTF-exposed plants 419 did not show changes in Car, in terms of their presence, quantity, and proportion (in 420 comparison with chlorophylls, assessed by the determination of TChl/Car ratio). The 421 422 ratio of TChl to Car is more informative that the simple measurement of pigment levels, because it indicates the degree of adaptation of plants to light and to adverse conditions. 423 In this case, after the evaluation of the obtained results, it is possible to suggest that the 424 425 here tested KTF concentrations were not capable to cause alterations of this physiological parameter, showing an adaptive capacity of plants when challenged by 426 this adverse condition. This assumption can be reinforced since, even after being 427 428 exposed to other NSAIDs, plants were capable to adapt to this source of chemical stress, and were not prone to suffer alterations in the Car content, along the entire duration of 429 430 the exposure (Alkimin et al., 2019a). This set of results concerning the levels of pigments demonstrates that L. minor was physiologically adapted to KTF, and no 431 deleterious effects were caused by KTF in terms of the pigments levels of exposed 432 433 organisms.

Finally, the observed biochemical effects were not evident enough to sustain the occurrence of physiological alterations. Despite the occurrence of biochemical changes caused by KTF exposure, these were not followed by deleterious changes in photosynthetic pigments (Chl a, b, total and Car). Thus, it is possible to assume that the antioxidant and biotransformation systems of *L. minor* were efficient enough to protect these plants from other injuries.

440

441 Daphnia magna

442 The genus *Daphnia* is considered as a dominant herbivorous in the zooplankton group, and animals from this genus play an important role in temperate freshwater ecosystems 443 as primary consumers (Smirnov, 2013). Thus, their feeding activity is an important 444 445 aspect in their ecology, being necessarily assessed in a stressful situation, including 446 exposure to contamination by pharmaceutical drugs. In this work, it was possible to observe that KTF, in the tested concentrations, did not alter feeding; animals subjected 447 448 to the recovery period evidenced a similar response, and no substantial effects were ascertained in these organisms. However, other NSAIDs, such as diclofenac, are 449 capable of reducing D. magna feeding rate (Nkoom et al., 2019) in concentrations 450 between 5 and 100  $\mu$ g/L. However, it is necessary consider that Nkoom et al. (2019) 451 experiment was conducted with neonates <24 h old, and our work involved only 452 453 juveniles (5 d old), which can be more tolerant (Hoang and Kaine, 2007). Feeding rate can affect a large number of life-history traits that may have context-dependent effects 454 on fitness (Garbutt and Little, 2014), being putatively associated with the reproduction. 455 456 Indeed, KTF did not affect the studied life-history traits. Data for another crustacean species, namely Ceriodaphnia dubia, showed the absence of effects after exposure to 457 KTF at concentrations up to  $100 \,\mu g/L$ , with a decrease in the offspring per female, when 458

animals were exposed to 1 mg/L (Mennillo et al., 2017). This level is nevertheless much higher than the here tested concentrations, and also well above the amounts detected in the environment, as already cited. These data are coherent, since the *D. magna* 48h  $EC_{50}$ , is higher than 100 mg/L of KTF (Boström and Berglund, 2015), a value approximately 3300 times higher than the levels tested in this study.

464 Despite not causing alterations in the feeding behavior and in life-history traits, KTF 465 was able to trigger the D. magna antioxidant system, increasing the CAT activity of 466 exposed animals. This occurred most likely as an attempt to cope with the putatively higher levels of generated ROS. Among NSAIDs, KTF has been shown to be more 467 468 toxic than other drugs from this class to the *D. magna* antioxidant system, generating changes in the antioxidant mechanisms in concentration as low as 0.24 µg/L of KTF 469 (943.8 pM). Daniel et al. (2019) reported an increase of CAT activity in animals 470 exposed to 2.56 mg/L (16.9 µM) of paracetamol, and, even a concentration of 50 µg/L 471 (242.4 nM) of ibuprofen was not able to trigger this system (Wang et al., 2016). 472

473 KTF seems also to have activated the biotransformation system in D. magna, by 474 increasing the activity of GSTs, a response that occurred for animals exposed to the intermediate concentrations (1.2 and 6  $\mu$ g/L). Phase II enzymes are crucial in 475 476 eliminating NSAIDs, but the interaction between GSTs and KTF is poorly understood for aquatic organisms. In other animals, such as mammals, KTF can be eliminated in its 477 free forms: conjugated via glucuronic acid, or as hydroxyl (OH) metabolite, with or 478 without conjugation (Alkatheeri et al., 1999). However, the metabolic pathway of 479 conjugation seems also to occur in aquatic organisms. In fact, KTF triggered the 480 biotransformation system, increasing the GSTs activity, in a fish PLHC-1 cell line 481 (Mennillo et al., 2017). Thus, it is possible to hypothesize that KTF can be excreted 482 following conjugation with glutathione (via GSTs) and/or with glucuronic acid, prior to 483

be delivered to phase III detoxification system. In addition, the lowest (0.24 µg/L) and 484 the highest (30 µg/L) concentrations did not cause any alteration of GST activity. Thus, 485 hermetic effect can be interpreted as follows: at low concentrations hydroxylation 486 processes were effective enough to detoxify the pharmaceutical, turning the phase II 487 biotransformation system of glutathione conjugation redundant (Bartha, 2012). 488 However, in animals exposed to higher concentrations, more complex metabolic 489 pathways should be activated by the drug (Bartha, 2012) leading to no alteration in the 490 491 GSTs activity. On the other hand, as already mentioned, ROS have the capacity to cause molecular damage and this effect can be linked to a decrease in enzymatic activity by 492 denaturation (Schieber and Chandel, 2014) and consequently, may be responsible for 493 the decrease of GSTs activity in the higher concentration, attaining values near those of 494 the control treatment. 495

Eicosanoids are oxygenated metabolites of arachidonic acid (AA) with different 496 functions. Their biosynthesis may occur according to different pathways, such as the 497 cytochrome P450 epoxygenase pathway (epoxyeicosatrienoic acids), lipoxygenage 498 499 (LOX) pathway (leukotrienes and lipoxins), and cyclooxygenase (COX) pathway (prostanoids: thromboxane and prostaglandins) (Stanley, 2006). As mentioned, COX or 500 501 PGH<sub>2</sub> synthase is an enzymatic form producing prostaglandins, and it is conserved in crustaceans, such as daphnids (Kyoto Encyclopedia of Genes and Genomes, KEGG). 502 Prostaglandins are signaling molecules with crucial importance in crustacean 503 504 reproduction and immune system (Smirnov, 2017). NSAIDs can interfere/inhibit 505 prostaglandins biosynthesis, since NSAIDs are competitive inhibitors of some of the involved enzymatic forms, by competing with AA, which is the physiological substrate 506 507 of COX; NSAIDs thereby affects eicosanoids biosynthesis and their physiological functions (Charlier and Michaux, 2003). Heckmann and co-workers (2008a) 508

demonstrated the existence of a COX pathway in D. magna, which appears to be 509 simpler than the COX pathway observed in mammals. In addition, another study 510 indicated that ibuprofen, also a NSAID, is capable of affecting the sequential processes 511 of oogenesis and embryogenesis in daphnids, by interrupting the COX metabolic 512 513 pathway of eicosanoids (Heckmann et al., 2008b). By acting this way, this drug is responsible for a dose-dependent decrease in *D. magna* reproduction (Han et al., 2010; 514 Heckmann et al., 2007) and also affects reproduction in another crustacean species, 515 516 namely Moina macrocopa (Han et al., 2010). The NSAID effect in COX was also observed in this study, since KTF exposure clearly decreased COX activity in a dose-517 response manner, as already mentioned before. This result was interpreted as a proxy 518 for the mode of action of these pharmaceutical drug; being a NSAIDs, its acute effects 519 resulted in the inhibition of COX, even at the low, albeit realistic here tested 520 521 concentrations, which were indisputably enough to trigger this mechanism in *D. magna*. However, reproductive effects were not observed in D. magna after KTF exposure, 522 523 which means that NSAIDs affect crustacean reproduction at higher concentrations (in 524 the mg/L order) as observed by Han et al. (2010) and Heckmann et al. (2007). These assumptions reinforce the notion that deleterious NSAIDs effects in the eicosanoids 525 biosynthesis pathway may be important for the reproduction of crustaceans. However, 526 527 the alterations that were observed in this study constitute a warning for the possible long-term effects of KTF in a worst environmental case scenario. 528

529

# 530 Conclusions

In conclusion, both species responded to low, realistic levels of KTF. However, the
toxic effects of KTF were only limited to biochemical parameters, without extrapolating
to physiological and population modifications. However, changes of the biochemical

534 parameters were clearly different between both species, being L. minor more susceptible to low KTF concentrations, hypothesizing the resilience capacity and hormetic effects; 535 while D. magna, in general, presented a dose response pattern. This comparison is 536 537 highly interesting and allows us to assume that KTF toxicity is eminently speciesspecific. Finally, the results here obtained are a clear indication about the possibility of 538 KTF in causing toxicological effects in the aquatic organisms, even when organisms 539 were exposed to realistic conditions simulating the already described scenarios of 540 541 contamination in the wild.

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838 Figure 1 – Effects of ketoprofen exposure in *L. minor* biomarkers upon exposure of 96 hours. A) 839 glutathione S-transferases activity; B) catalase activity; and C) carbonic anhydrase activity. Bars and 840 errors bars are Mean  $\pm$  SE (N= 10). \* stands for statistical differences in relation to control following 841 ANOVA and Dunnett's or the equivalent non-parametric test.



Figure 2 – Effects of ketoprofen exposure in *L. minor* during 96 hours. A) chlorophyll a content, B) chlorophyll b content, C) total chlorophyll content, D) carotenoids content, E) chlorophyll a and b ratio and F) total chlorophyll and carotenoids ratio. Bars and errors bars are Mean  $\pm$  SE (N= 10).



847 Figure 3 – Effects of ketoprofen in *D. magna* after acute exposure (48h). A) catalase activity B)

848 glutathione S-transferases activity, and C) cyclooxygenase activity. Bars and errors bars are Mean  $\pm$  SE 849 (N= 5). \* stands for statistical differences in relation to control following ANOVA and Dunnett's or the 850 equivalent non-parametric test.



852 Figure 4 - Effects of ketoprofen in D. magna feeding behavior, after exposure and after recovery (post-853 exposure). Bars and errors bars are Mean  $\pm$  SE (N= 5).

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Figure 5 – Ketoprofen effects in reproductive parameters in *Daphnia magna*. A) day of the first brood, B) number of neonates from the first brood and C) total number of neonates (three broods). Bars and errors bars are Mean  $\pm$  SE (N= 10).

# Highlights

- ✓ Ketoprofen (KTF) acts different on *Lemna minor* and on *Daphnia magna*;
- ✓ Effects of KTF were limited only to biochemical effects on both species;
- ✓ L. minor effects of low KTF concentrations suggest resilience and hormetic effects;
- ✓ D. magna, in general, presented a dose response pattern to KTF exposure.

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### **Authors statement**

GDA – Conceptualization; Methodology; Formal analysis; Investigation; Visualization;Writing - Original Draft; Writing - Review & Editing.

AMVMS Writing - Review & Editing; Supervision; Project administration; Funding acquisition

CB - Writing - Review & Editing; Supervision; Project administration; Funding acquisition

BN – Visualization; Writing - Review & Editing; Supervision; Project administration; Funding acquisition

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### **Declaration of interests**

 $\boxtimes$  The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

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