





Exploring the high-temperature thermoelectric performance of Al-doped ZnO ceramics prepared by in-situ aluminothermic reactions

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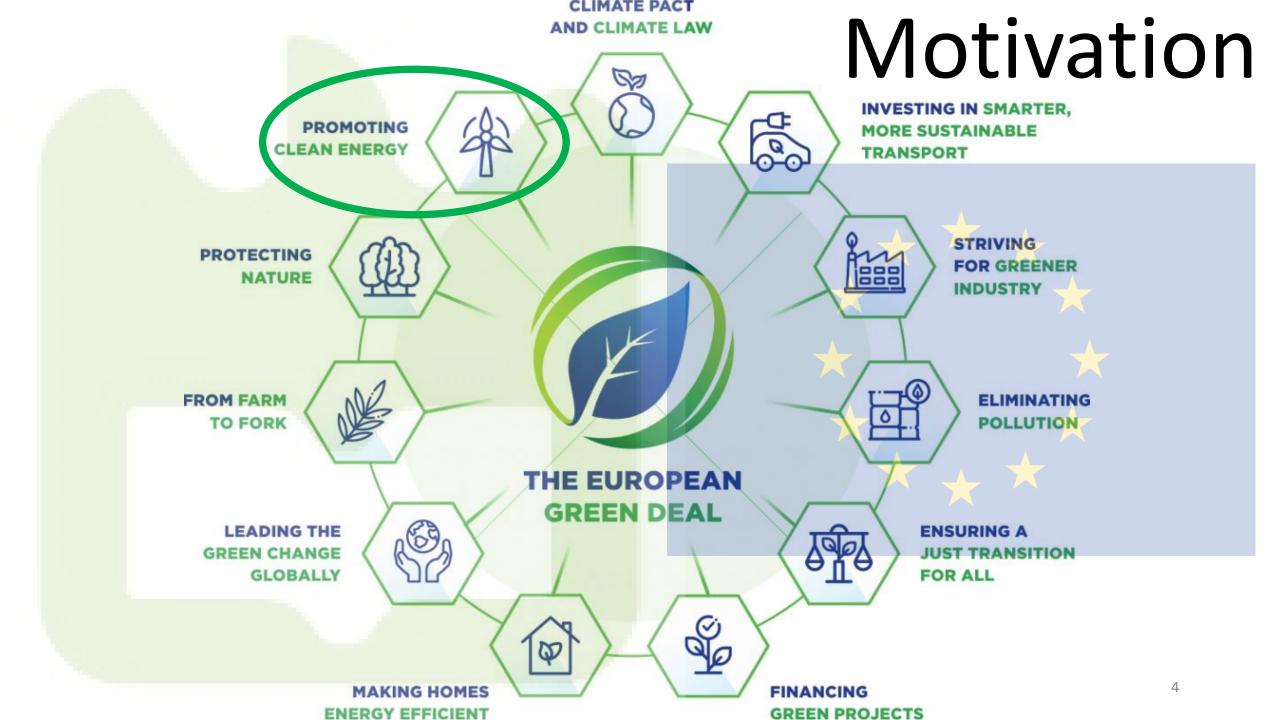


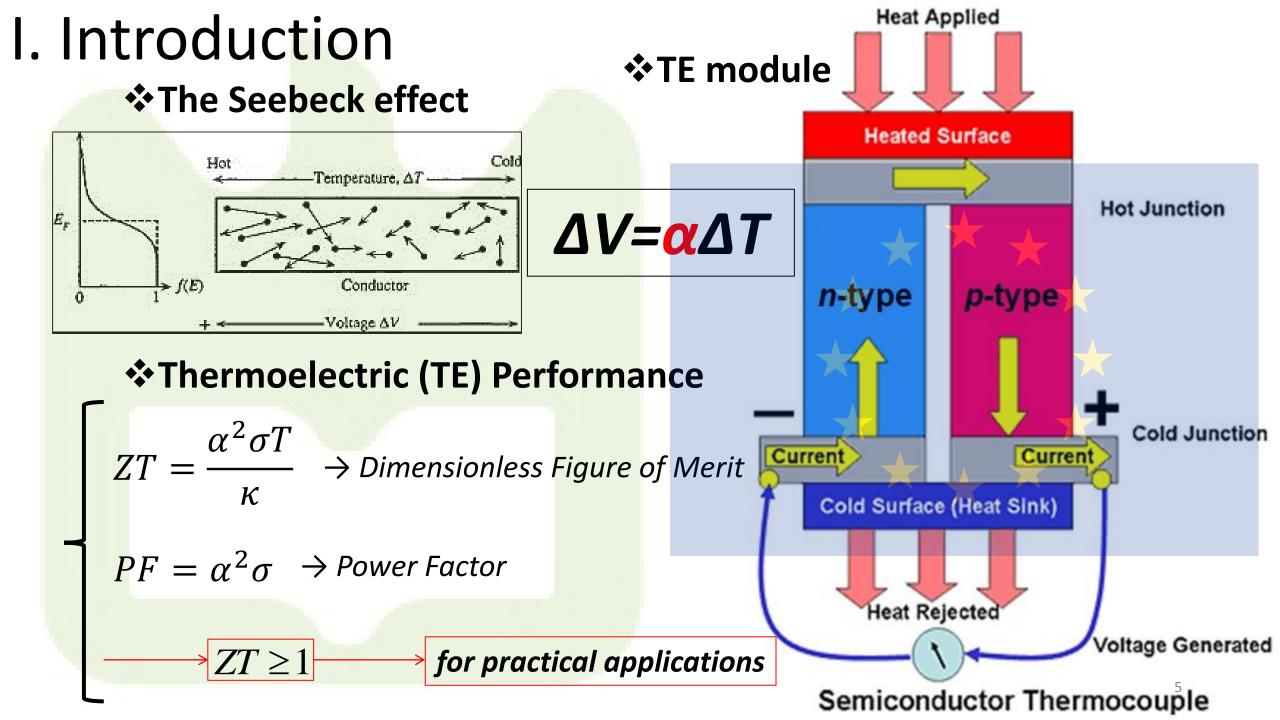
Outline:

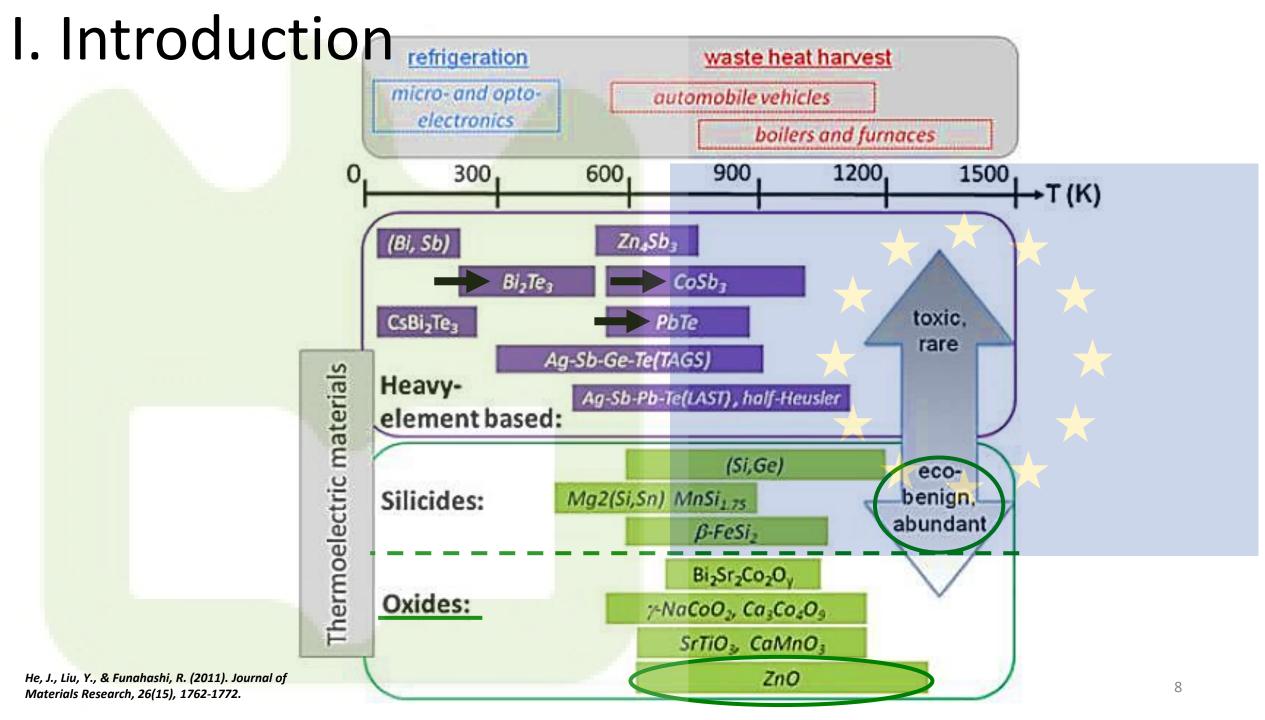
I. Introduction

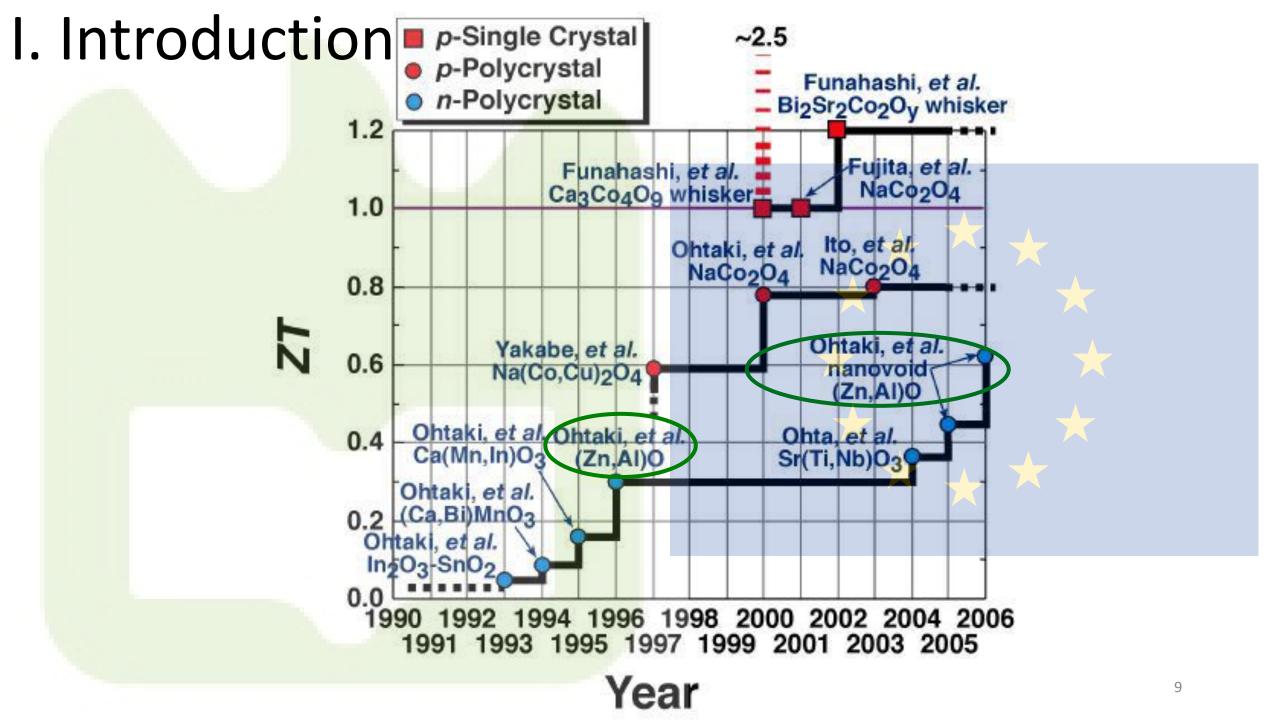
II. Experimental details III. Preliminary results IV. Conclusions

V. Ongoing work









I. Introduction

Wurtzite structure

7n²⁺

- * Aluminothermy: a process of producing great *heat and strong chemical reduction by oxidizing* finely divided aluminum with oxygen taken from another metal, this metal being thus reduced from its oxide (as molten iron is obtained from *iron oxide in welding by the Thermit process)* Displacement reaction between Al and ZnO [1]: $2Al + 3ZnO \rightarrow Al2O3 + 3Zn + \Delta H$, where the enthalpy $\Delta H_T = \Delta H_0 + \int \Delta C_p dT$ [2]
- ✓ n-type conduction; direct wide bandgap (~3.37 eV); low *n* and large κ ; relatively high electronegativity; high μ is expected due to strong preference towards *sp*³ hybridization

[1] 10.1007/s10853-010-4619-9; [2] 10.1016/S0921-5093(97)00544-3

II. Experimental details

Synthesis: Solid-state reaction (targeting Zn_{0.995}Al_{0.005}O (ZAO)) and uniaxial pressing at 227 MPa

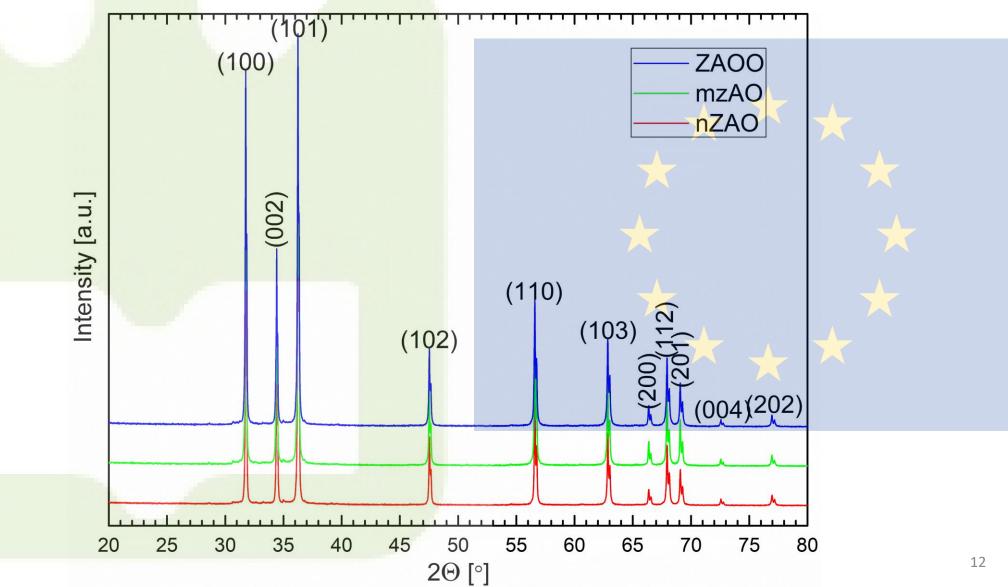
Samples:ZnO + $<math display="block"> \frac{\text{Inanometric Al powder}}{\text{Inanometric Al powder}} = \frac{\text{InZAO}}{\text{InZAO}}$

Sintering: One step sintering in air (RT-5°C/min-1300°C,10h-5°C/min-RT)

Composition, morphology and microstructure: XRD, SEM-EDS

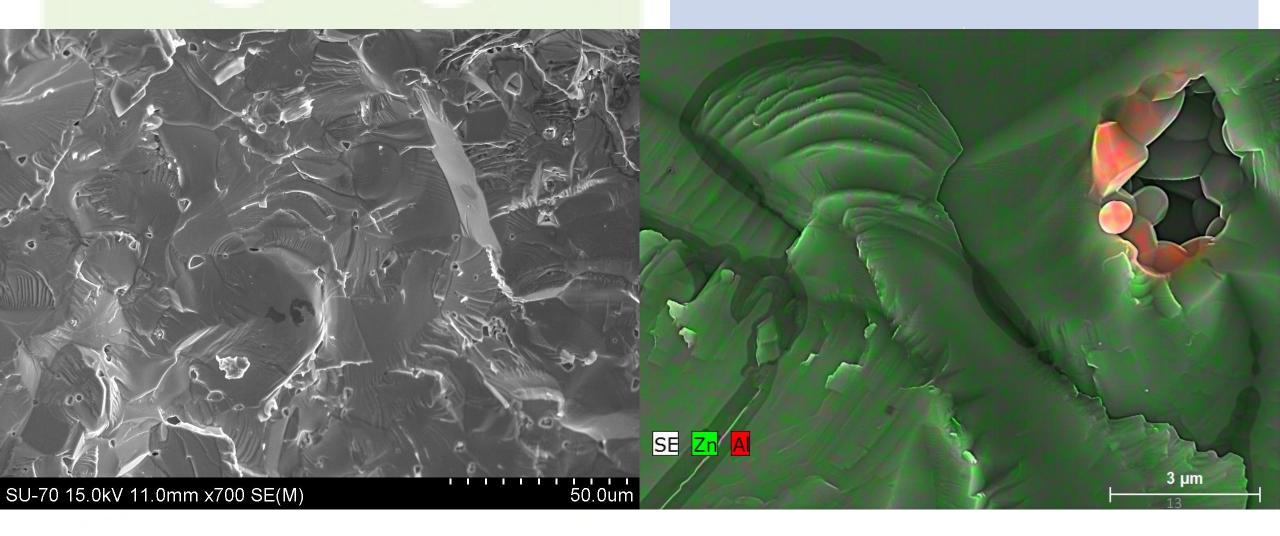
***** Electrical properties: σ , α , PF (steady state, 525 – 1175 K, 50K/30' step, air)

100% Hexagonal Wurtzite ZnO; NO ZnAl₂O₄ spinel detected



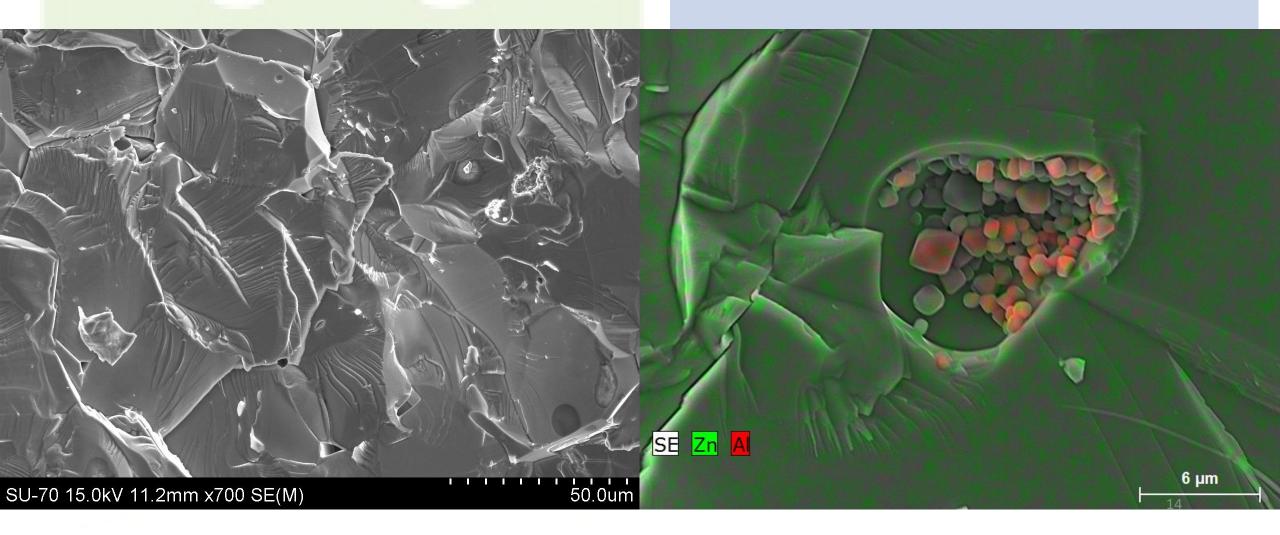
✤nZAO

 $\rho_{exp.}$ =5.46g/cm³ (96% of $\rho_{th.}$)



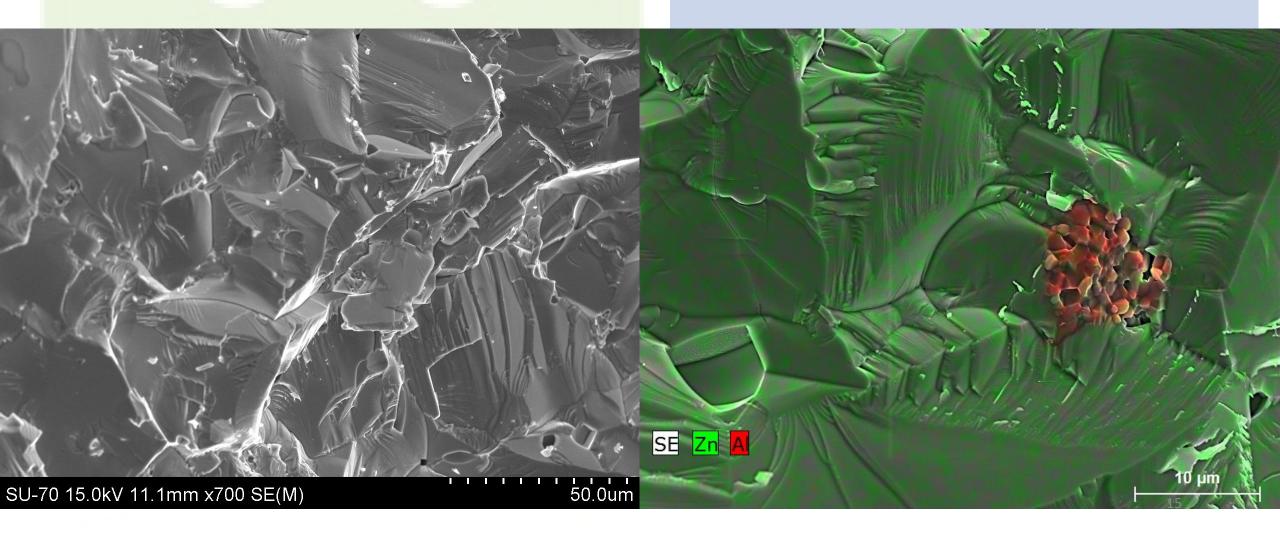
✤mZAO

 $\rho_{exp.}$ =5.44g/cm³ (96% of $\rho_{th.}$)

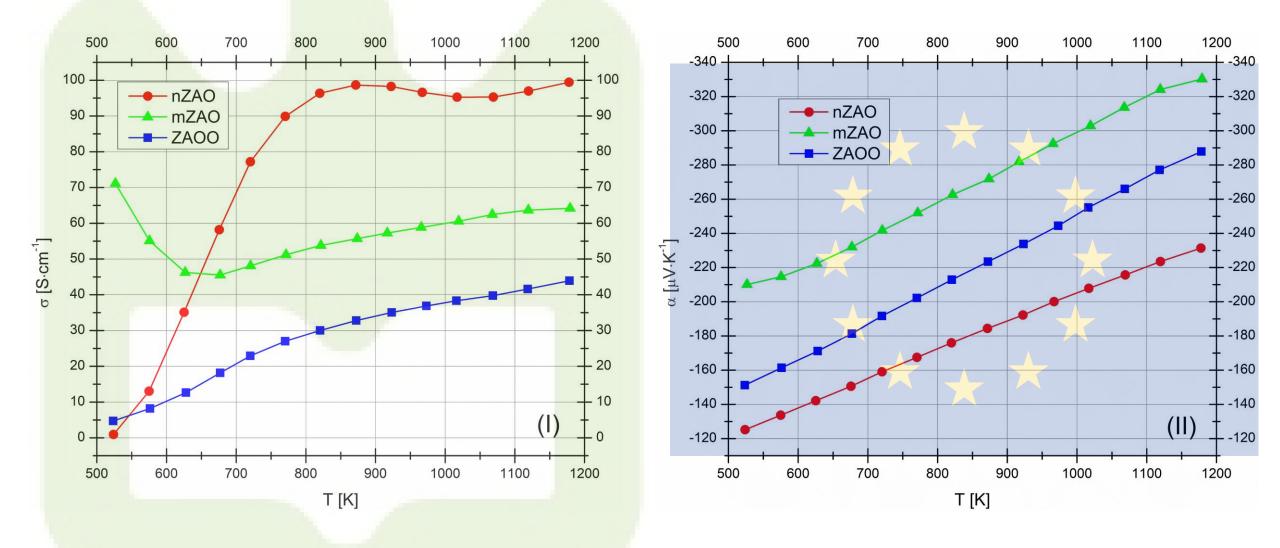


∜ZAOO

 $\rho_{exp.}$ =5.23g/cm³ (92% of $\rho_{th.}$)



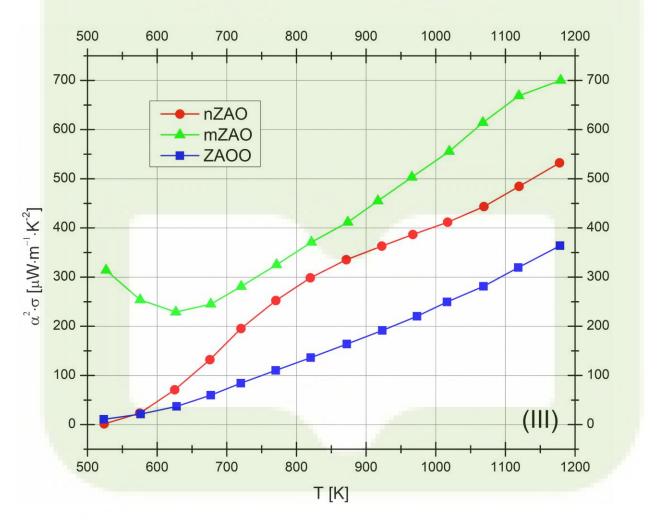
Sigma and alpha*

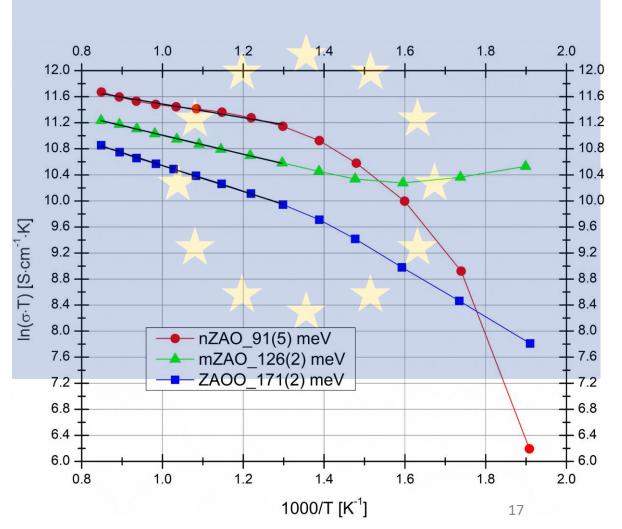


*The estimated experimental error in measured values did not exceed 3-5% for σ and 5-7% for α .

 $\mathbf{PF} \& E_{a}(\sigma)$

Ref. 10.1039/c8ta01463a: ~300 μWm⁻¹K⁻² at 900 °C, for Zn_{0.993}Al_{0.007}O samples prepared via solid state reaction





IV. Conclusions

- High-density (>95%), single-phase Zn_{0.995}Al_{0.005}O (ZAO) ceramic composites have been successfully prepared from 2 different metallic Al powders, through a classical solid-state reaction involving high-temperature, in-situ aluminothermy
- The nZAO and mZAO samples show improved properties of interest (density, electrical performance), compared to the reference ZAOO samples
- The highest incorporation of Al is achieved for the nanometric Al powders
- * The highest PF recorded for mZAO (700 μWm⁻¹K⁻²) is among the best found in literature, for similar TE compositions
- In-situ aluminothermy provides locally-strong redox conditions which facilitate sintering and allows for an efficient tuning of the microstructural design and the high-temperature TE properties of these novel ZnO-based materials

V. Ongoing work

* Thermal diffusivity D and specific heat capacity C_p studies performed in similar

conditions as for the electrical measurements ($\rightarrow \kappa = D\rho C_p$; $\kappa_{ph} = \kappa - \sigma LT$)

- * Rietveld refinement and UV-Vis Diffuse Reflectance Spectroscopy (DRS) for determination of unit cell parameters and optical band-gap energy E_q , respectively
- Charge carrier concentration and/or mobility calculations
- Experimentation with multiple annealing steps and shorter sintering schemes (with T > 1000 °C), for a further evaluation/design/tuning of the redox-promoted, in-situ aluminothermy approach
- Stability experiments (time dependences of electrical coefficients) in various conditions of temperature and atmosphere
 ¹⁹

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(Some) Applications of Thermoelectricity:

Power (Watts)

10k

Ik

100

10

100m

10m

lm

100µ

10µ

Large scale waste heat recovery, e.g., transportation vehicles

Microelectronics, e.g., CPU cooling

Consumer applications, e.g., Wine coolers

Space power and cooling, e.g., Voyager

Remote wireless sensors, e.g., networks

Biomedical devices, e.g., pacemakers

Low-power applications, e.g., wrist watches

